

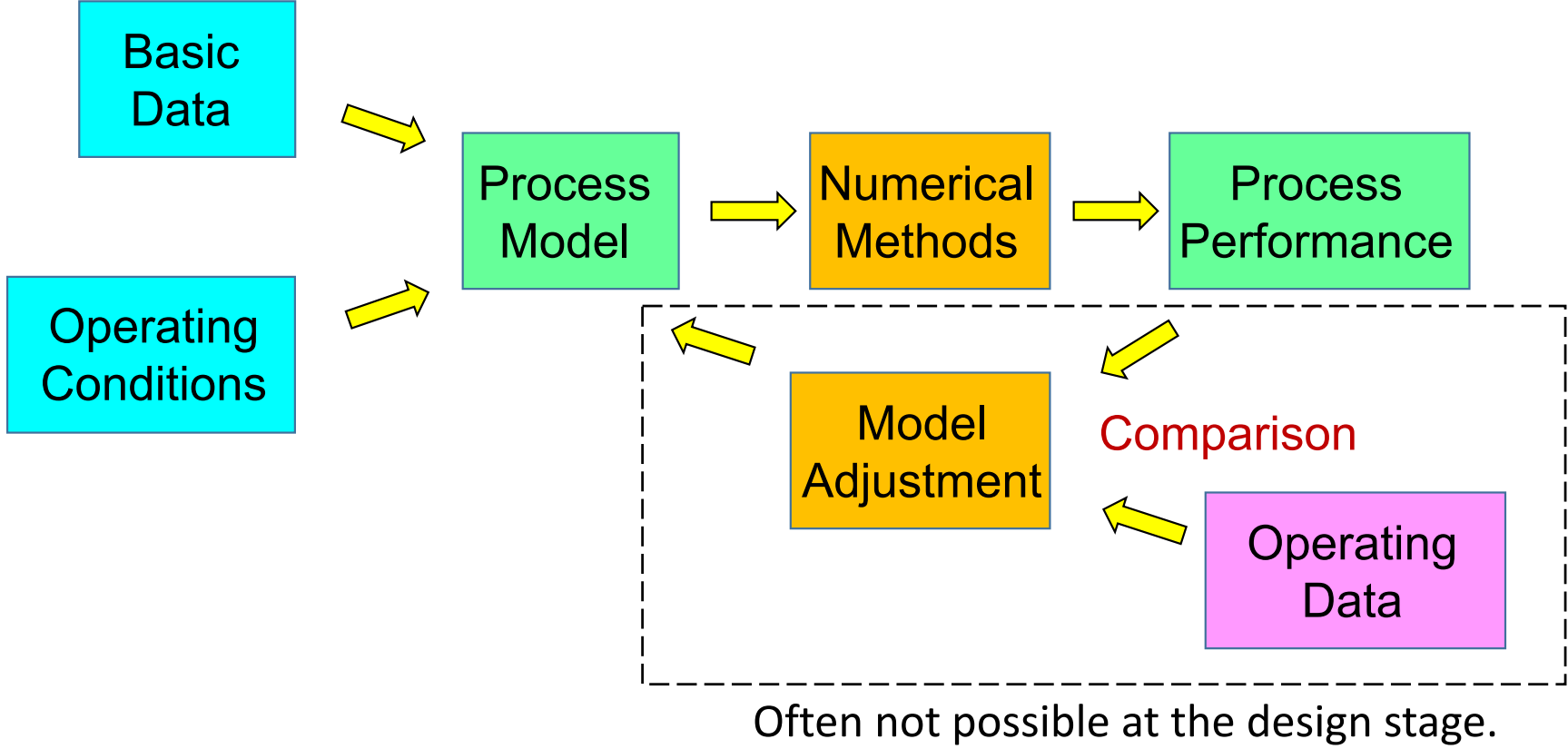
## Python Bootcamps 1, 2 and 3

- ✓ 1: Getting up to speed with Python
- 2: Learning to use Python to solve typical problem scenarios
- 3: Detailed modeling of packed-bed and plug-flow reactors

### Bootcamp 2 Outline

- Process modeling & numerical characteristics
- Algebraic models
  - Single, nonlinear, including polynomials
  - Linear sets
  - Nonlinear sets
- ODE models
  - Initial value problems
  - Split boundary problems
- Optimization
- Curve-fitting

# Process Modeling



# Process Modeling

## Developing the Process Model

### Conservation Balances

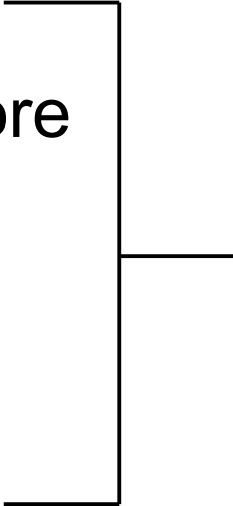
- Material
- Energy
  - Mechanical/Momentum
  - Thermal
- Thermodynamics
  - Equilibrium, Phase and Chemical
  - Heat Transfer
- Mass Transfer
- Separations
- Reaction Kinetics
- Monitoring and Control

## Equipment

- Vessels
  - Tanks, Drums
  - Columns
  - Reactors
- Heat Exchangers
- Piping, Valves, Fittings
- Pumps, Compressors
- Columns
- Reactors
- Solids-handling
  - Crystallization
  - Filtration
- Instrumentation

# Numerical Characteristics

- Algebraic equations
  - Nonlinear, one or more
  - Linear sets
- Differential equations
  - Ordinary (ODEs)
  - Partial (PDEs)
- Optimization
- Curve-fitting



when combined:  
Differential-Algebraic  
Systems (DAEs)

# Examples Considered

- Single, nonlinear algebraic equation
  - water-gas shift equilibrium
- Set of linear algebraic equations
  - absorber column
- Set of nonlinear algebraic equations
  - steam/water equilibrium
- Single nonlinear ordinary differential equation
  - batch reactor, single reaction
- Set of nonlinear ordinary differential equations
  - batch reactor, multiple reactions
- Optimization
  - single factor
  - multiple factors with constraints
- Linear Regression
  - polynomial
  - general

# Solving Single Algebraic Equations

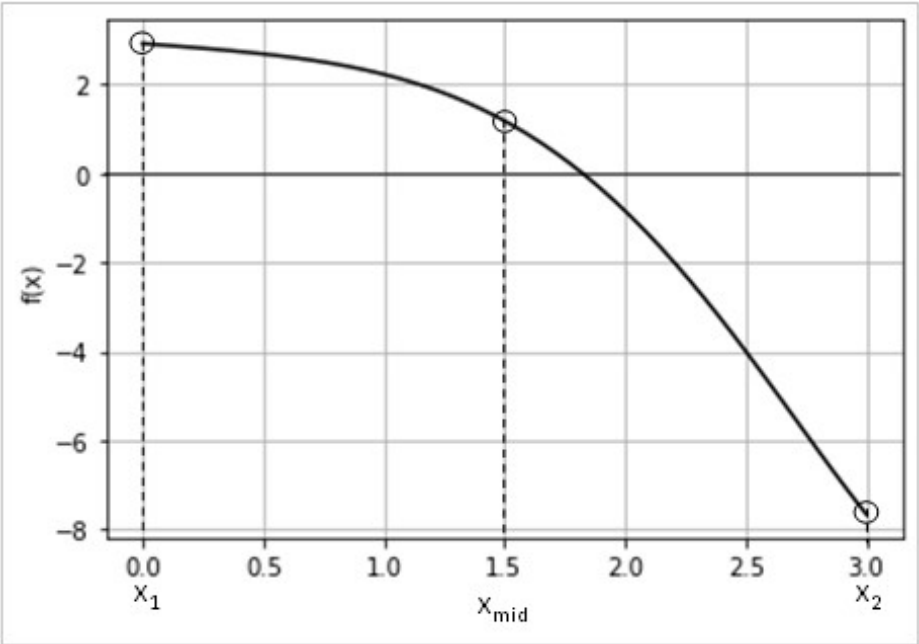
- Bracketing methods
  - Bisection
  - False position
- Open methods
  - Newton-Raphson
  - Modified secant
- Hybrid
  - Brent's method
- Circular scenario
  - Substitution
  - Wegstein method

$$f(x) = 0$$

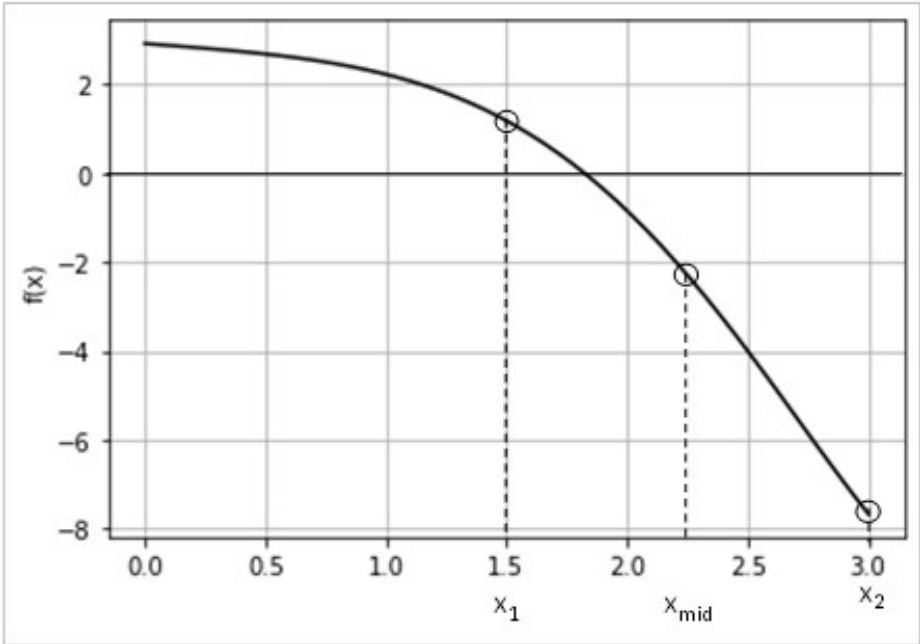
$$x = g(x)$$

For details and Python code on all these methods, see Chapra and Clough, *Applied Numerical Methods with Python for Engineers and Scientists*, McGraw-Hill, 2022. <sup>6</sup>

# Solving Single Algebraic Equations - Bisection



First iteration



Second iteration

# Solving an Algebraic Equation with Bisection

```
def bisect(func,x1,x2,maxit=20):
    """
    Uses the bisection method to estimate a root of func(x).
    The method is iterated maxit (default = 20) times.
    Input:
        func = name of the function
        x1 = lower guess
        x2 = upper guess
    Output:
        xmid = root estimate
        or
        error message if initial guesses do not bracket solution
    """
    if func(x1)*func(x2)>0:
        return 'initial estimates do not bracket solution'
    for i in range(maxit): # carry out maxit iterations
        xmid = (x1+x2)/2 # calculate midpoint
        if func(xmid)*func(x1)>0: # check if f(xmid) same sign as f(x1)
            x1 = xmid # if so, replace x1 with xmid
        else:
            x2 = xmid # if not, replace x2 with xmid
    return xmid # return the latest value of xmid as the solution
```

docustring

**bisect1.py**



# Solving an Algebraic Equation with Bisection

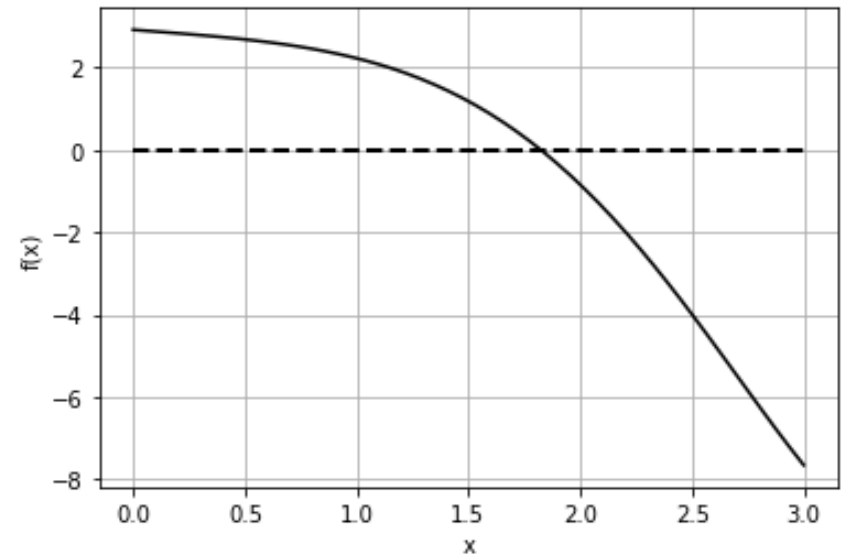
Example  $f(x) = \sin(x+2) \cdot \cosh(x) + 2 = 0$   $x_1 = 0$   $x_2 = 3$

```
import numpy as np

def f(x):
    return np.sin(x+2)*np.cosh(x)+2.

x1 = 0. ; x2 = 3.
xsoln = bisect(f,x1,x2)
print('solution is {0:6.4f}'.format(xsoln))
```

solution is 1.8229



# Solving an algebraic equation with SciPy brentq function from the optimize submodule

`xs = brentq(f,x1,x2)`     `f` : function name  
                          `x1,x2` : initial estimates that bracket the solution

Example      $f(x) = \sin(x+2) \cdot \cosh(x) + 2 = 0$       $x_1 = 0$       $x_2 = 3$

```
import numpy as np
from scipy.optimize import brentq

def f(x):
    return np.sin(x+2)*np.cosh(x)+2.
```

```
x1 = 0.
x2 = 3.
```

solution is 1.8229

```
xsoln = brentq(f,x1,x2)
```

```
print('solution is {0:6.4f}'.format(xsoln))
```

**brent1.py**

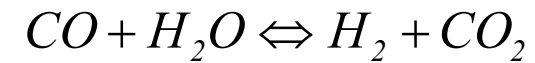
# Solving an algebraic equation with SciPy brentq function from the optimize submodule

full output option

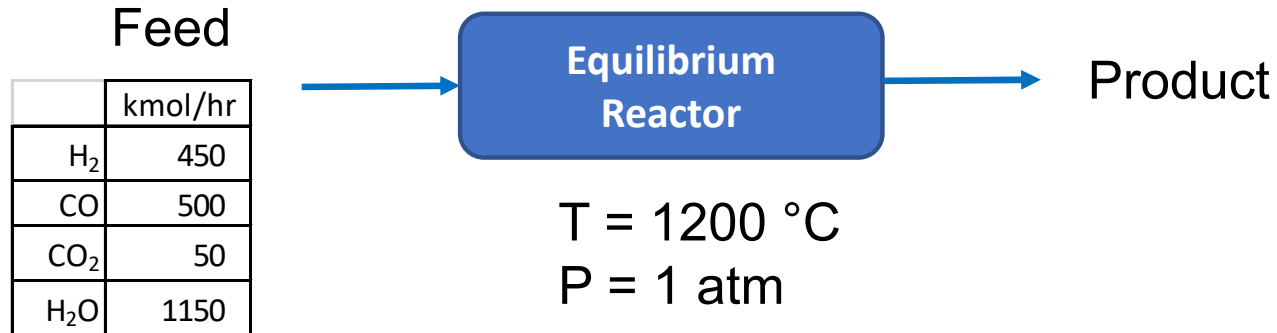
```
xsoln,results = brentq(f,x1,x2,full_output=True)
print('solution is {0:6.4f}'.format(xsoln))
print(results)
```

```
solution is 1.8229
      converged: True
      flag: 'converged'
function_calls: 10
iterations: 9
      root: 1.8228849971055754
```

## Example: water-gas shift equilibrium



$$\frac{[H_2] \cdot [CO_2]}{[H_2O] \cdot [CO]} = K_{eq}(T) \quad \ln[K_{eq}(T)] = -3.112 + \frac{3317}{T} \quad T(K)$$



$$f(x) = \frac{[Feed_{H_2} + x] \cdot [Feed_{CO_2} + x]}{[Feed_{H_2O} - x] \cdot [Feed_{CO} - x]} - K_{eq}(T) = 0$$

where  $x$  is the shift to equilibrium in kmol/hr.

Solve for  $x$  and the reactor product flow rates for the given temperature.

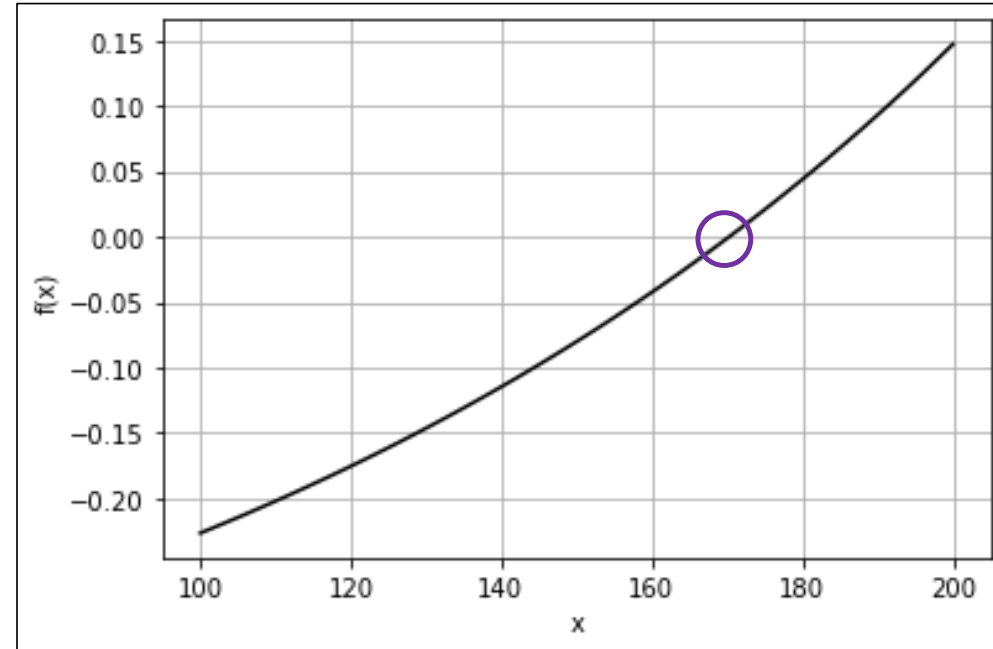
# Water-gas shift equilibrium

Setting up a function to compute  $f(x)$

```
import numpy as np
import matplotlib.pyplot as plt

def f(x):
    T = 1200 # degC
    TK = T + 273.15 # K
    Keq = np.exp(-3.112+3317/TK) # equil constant
    FeedH2 = 450 # kmol/hr
    FeedCO2 = 50
    FeedH2O = 1150
    FeedCO = 500
    ferr = (FeedH2+x)*(FeedCO2+x)/(FeedH2O-x)/(FeedCO-x) - Keq
    return ferr
```

```
x = np.linspace(100,200)
plt.plot(x,f(x),c='k')
plt.grid()
plt.xlabel('x')
plt.ylabel('f(x)')
```



Plot shows solution  $x = \sim 170$

**watargasplot.py**

# Water-gas shift equilibrium

Setting up a function to compute  $f(x)$   
including temperature and feed rates as arguments

```
def f(x, T, FeedH2, FeedCO2, FeedH2O, FeedCO):  
    TK = T + 273.15 # K  
    Keq = np.exp(-3.112+3317/TK) # equil constant  
    ferr = (FeedH2+x)*(FeedCO2+x)/(FeedH2O-x)/(FeedCO-x) - Keq
```

Case study of equilibrium flow rates for a range of temperatures

# Water-gas shift equilibrium

Passing extra arguments through a built-in function

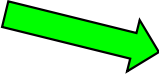
Main script

Set T and Feed rates

Set x1 and x2



```
xsoln = brentq(f,x1,x2,args=(...))
```



function f with arguments  
x, T, Feed rates

need to get these values “through”  
brentq to f using  
args=(...)

# Water-gas shift equilibrium

Case study for a range of temperatures with a plot

```
import numpy as np
import matplotlib.pyplot as plt
from scipy.optimize import brentq

def f(x,T,FeedH2,FeedCO2,FeedH2O,FeedCO):
    TK = T + 273.15 # K
    Keq = np.exp(-3.112+3317/TK) # equil constant
    ferr = (FeedH2+x)*(FeedCO2+x)/(FeedH2O-x)/(FeedCO-x) - Keq
    return ferr

FeedH2 = 450 # kmol/hr
FeedCO2 = 50
FeedH2O = 1150
FeedCO = 500

T = np.arange(500,1525,25)
n = len(T)
x = []
ProdH2 = []
ProdCO2 = []
ProdH2O = []
ProdCO = []
```

watergas\_solve.py



Use NumPy **append** function here to extend arrays instead of creating zero-filled arrays.



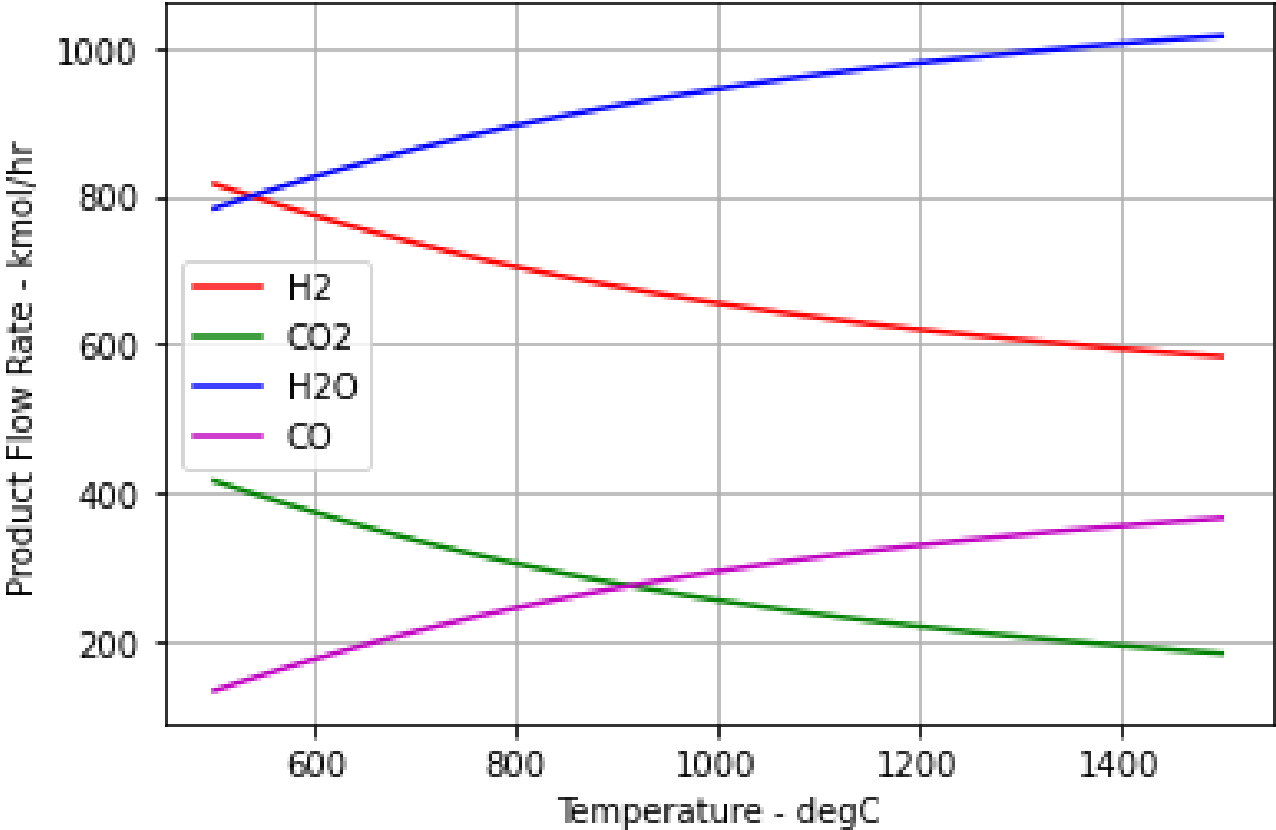
```
x1 = 50 ; x2 = 450
for i in range(n):
    xsoln = brentq(f,x1,x2,args=(T[i], \
        FeedH2,FeedCO2,FeedH2O,FeedCO))
    x = np.append(x,xsoln)
    ProdH2 = np.append(ProdH2,FeedH2+xsoln)
    ProdCO2 = np.append(ProdCO2,FeedCO2+xsoln)
    ProdH2O = np.append(ProdH2O,FeedH2O-xsoln)
    ProdCO = np.append(ProdCO,FeedCO-xsoln)

plt.plot(T,ProdH2,c='r',label='H2')
plt.plot(T,ProdCO2,c='g',label='CO2')
plt.plot(T,ProdH2O,c='b',label='H2O')
plt.plot(T,ProdCO,c='m',label='CO')
plt.grid()
plt.legend()
plt.xlabel('Temperature - degC')
plt.ylabel('Product Flow Rate - kmol/hr')
```



# Water-gas shift equilibrium

Case study for a range of temperatures with a plot



# Solving for the Roots of Polynomials

General form for  $n^{\text{th}}$ -order polynomial

$$a_n x^n + a_{n-1} x^{n-1} + \dots + a_1 x + a_0 = 0$$

will have  $n$  roots, either real or complex conjugate pairs,  $\alpha + \beta j$   $\alpha - \beta j$

Example:  $x^4 - 8x^3 - 3x^2 + 62x + 56 = 0$

```
import numpy as np
coeff = [1., -8., -3., 62., 56.]
r = np.roots(coeff)
print('roots are\n',r)
```

roots are  
[ 7. 4. -2. -1.]

NumPy function **roots**

Example:  $x^2 + x + 1 = 0$

```
coeff2 = [1., 1., 1.]
r2 = np.roots(coeff2)
print('\nroots are\n',r2)
```

roots are  
[-0.5+0.8660254j -0.5-0.8660254j]

**polynomial\_example.py**

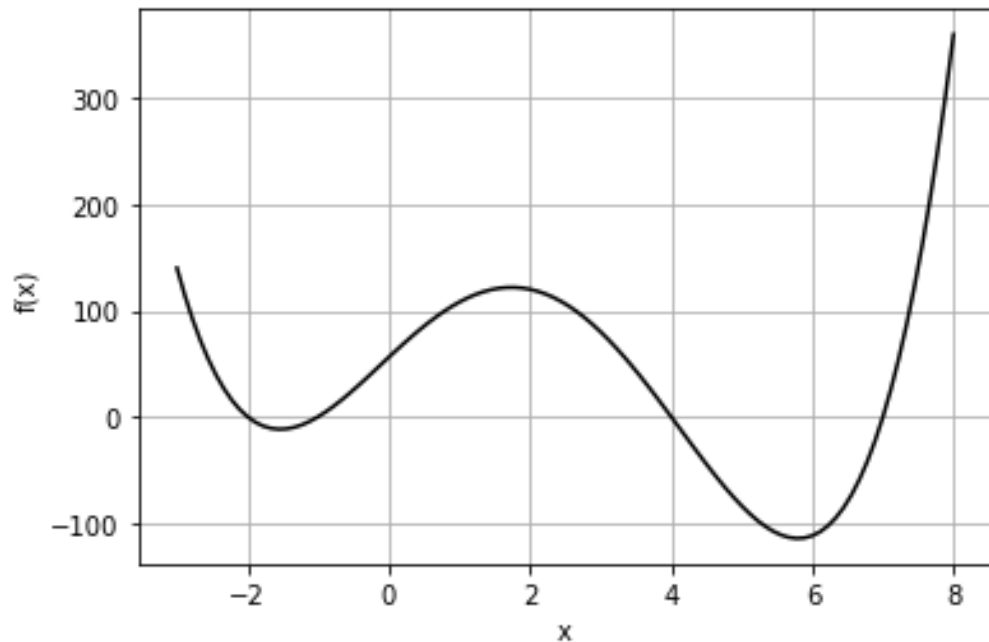
# Solving for the Roots of Polynomials

Composing the polynomial from its roots, NumPy **poly** function

```
coeff1 = np.poly([7., 4., -2., -1.])    coefficients are  
print('\ncoefficients are\n',coeff1)    [ 1. -8. -3. 62. 56.]
```

Evaluating a polynomial to create a plot

```
import numpy as np  
import matplotlib.pyplot as plt  
  
coeff = [1., -8., -3., 62., 56.]  
x = np.linspace(-3.,8.,100)  
y = np.polyval(coeff,x)  
  
plt.plot(x,y,c='k')  
plt.grid()  
plt.xlabel('x')  
plt.ylabel('f(x)')
```



**polynomial\_example2.py**

# Solving Sets of Linear Algebraic Equations

$n$  equations in  $n$  unknowns

$$\begin{aligned} a_{11}x_1 + a_{12}x_2 + \cdots + a_{1n}x_n &= b_1 \\ a_{21}x_1 + a_{22}x_2 + \cdots + a_{2n}x_n &= b_2 \\ &\vdots \\ a_{n1}x_1 + a_{n2}x_2 + \cdots + a_{nn}x_n &= b_n \end{aligned} \quad \begin{bmatrix} a_{11} & a_{12} & \cdots & a_{1n} \\ a_{21} & a_{22} & \cdots & a_{2n} \\ \vdots & \vdots & \ddots & \vdots \\ a_{n1} & a_{n2} & \cdots & a_{nn} \end{bmatrix} \cdot \begin{bmatrix} x_1 \\ x_2 \\ \vdots \\ x_n \end{bmatrix} = \begin{bmatrix} b_1 \\ b_2 \\ \vdots \\ b_n \end{bmatrix} \quad \mathbf{A} \cdot \mathbf{x} = \mathbf{b}$$

Solving the equations:

1. matrix algebra and computations  $\mathbf{A}^{-1} \cdot \mathbf{A} \cdot \mathbf{x} = \mathbf{A}^{-1} \cdot \mathbf{b}$

$$\mathbf{I} \cdot \mathbf{x} = \mathbf{A}^{-1} \cdot \mathbf{b}$$

$$\mathbf{x} = \mathbf{A}^{-1} \cdot \mathbf{b}$$

compute the inverse of  $\mathbf{A}$   
and multiply it by  $\mathbf{b}$

2. more efficient numerical method

- Gaussian elimination with enhancements
- LU decomposition

# Solving Sets of Linear Algebraic Equations

Example

$$3x + 2y - z = 10$$

$$-x + 3y + 2z = 5$$

$$x - y - z = -1$$

$$\begin{bmatrix} 3 & 2 & -1 \\ -1 & 3 & 2 \\ 1 & -1 & -1 \end{bmatrix} \cdot \begin{bmatrix} x \\ y \\ z \end{bmatrix} = \begin{bmatrix} 10 \\ 5 \\ -1 \end{bmatrix}$$

$\mathbf{A}^{-1} \cdot \mathbf{b}$   $\Rightarrow$  not recommended, inefficient

`simple_linear_set_linalg_solve.py`

```
import numpy as np
```

```
A = [[3, 2, -1], [-1, 3, 2], [1, -1, -1]]
```

```
b = [10, 5, -1]
```

```
x = np.linalg.solve(A,b)
```

```
print(x)
```

```
[-2.  5. -6.]
```

using the **solve** function  
in the NumPy **linalg** submodule

The **solve** function is adapted from the LAPACK package **gesv** routine and uses LU decomposition with partial pivoting.

# Solving Sets of Linear Algebraic Equations

Example problem: Six-stage absorber column

Equilibrium relationship  
on tray  $i$   $y_i = ax_i + b$

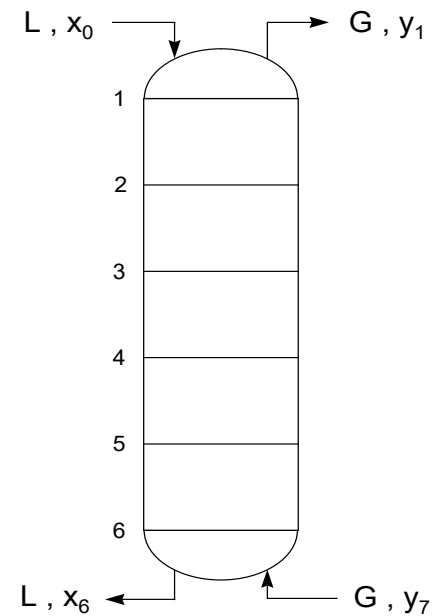
$x_0$ ,  $y_7$ ,  $L$  and  $G$  specified

Component material balance on tray  $i$

$$L \cdot x_{i-1} + G \cdot y_{i+1} = L \cdot x_i + G \cdot y_i$$

Incorporate equilibrium relationship

$$L \cdot x_{i-1} - (L + G \cdot a) \cdot x_i + G \cdot a \cdot x_{i+1} = 0$$



# Solving Sets of Linear Algebraic Equations

## Example problem: Six-stage absorber column

Write component material balances for each tray and rearrange with unknowns on the left and knowns on the right.

$$\begin{aligned}-(L + Ga)x_1 + Gax_2 &= -Lx_0 \\Lx_1 - (L + Ga)x_2 + Gax_3 &= 0 \\Lx_2 - (L + Ga)x_3 + Gax_4 &= 0 \\Lx_3 - (L + Ga)x_4 + Gax_5 &= 0 \\Lx_4 - (L + Ga)x_5 + Gax_6 &= 0 \\Lx_5 - (L + Ga)x_6 &= -G(y_7 - b)\end{aligned}$$

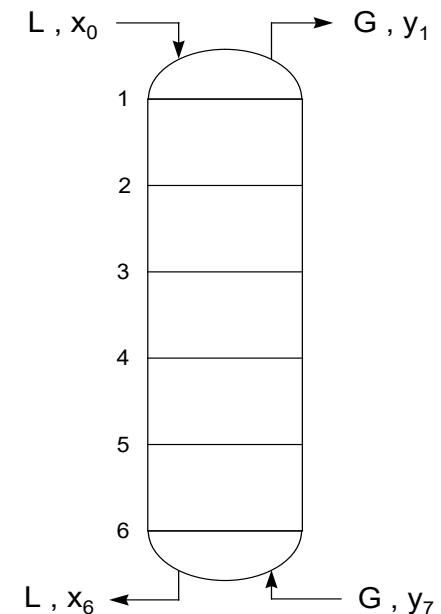
This represents a set of six linear equations in the six unknown mass fractions.

Basic data: equilibrium model:  $a = 0.7$ ,  $b = 0$

Operating conditions:  $L = 20$  mol/s,  $G = 12$  mol/s

Inlet gas mole fraction:  $y_7 = 0.1$

Inlet liquid mole fraction:  $x_0 = 0$



# Solving Sets of Linear Algebraic Equations

Example problem: Six-stage absorber column

```
import numpy as np
import matplotlib.pyplot as plt

a = 0.7 # equilibrium factor
yin = 0.1 # vapor entry mole fraction
L = 20 # mol/s
G = 12 # mol/s
n = 6 # number of stages
```

```
A = np.zeros((n,n))
b = np.zeros(n)
```

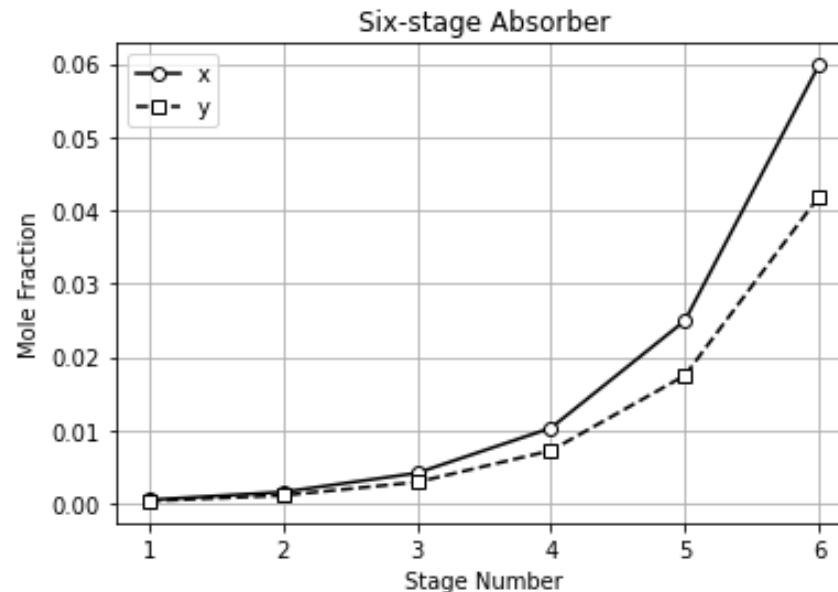
```
for i in range(n):
    A[i,i] = -(L+G*a)
for i in range(n-1):
    A[i,i+1] = G*a
for i in range(1,n):
    A[i,i-1] = L
b[5] = -G*yin
```

```
x = np.linalg.solve(A,b)
y = a*x
```

```
stage = np.arange(1,7)

plt.plot(stage,x,c='k',marker='o',mec='k',mfc='w',label='x')
plt.plot(stage,y,c='k',ls='--',marker='s',mec='k',mfc='w',label='y')
plt.grid()
plt.legend()
plt.xlabel('Stage Number')
plt.ylabel('Mole Fraction')
plt.title('Six-stage Absorber')
```

**Absorber.py**





# Solving Sets of Nonlinear Algebraic Equations

$$\begin{aligned} f_1(x_1, x_2, \dots, x_n) &= 0 \\ f_2(x_1, x_2, \dots, x_n) &= 0 \\ &\vdots \\ f_n(x_1, x_2, \dots, x_n) &= 0 \end{aligned} \quad \text{or} \quad \mathbf{f}(\mathbf{x}) = \mathbf{0}$$

Common solution technique: Newton's Method

Start with an initial estimate of the solution:  $\mathbf{x}^0$

Iterate with  $\mathbf{x}^{i+1} = \mathbf{x}^i - \mathbf{J}^{-1}(\mathbf{x}^i) \cdot \mathbf{f}(\mathbf{x}^i)$  until a convergence criterion is met.

Jacobian matrix

$$\mathbf{J} = \begin{bmatrix} \frac{\partial f_1}{\partial x_1} & \frac{\partial f_1}{\partial x_2} & \dots & \frac{\partial f_1}{\partial x_n} \\ \frac{\partial f_2}{\partial x_1} & \frac{\partial f_2}{\partial x_2} & \dots & \frac{\partial f_2}{\partial x_n} \\ \vdots & \vdots & \ddots & \vdots \\ \frac{\partial f_n}{\partial x_1} & \frac{\partial f_n}{\partial x_2} & \dots & \frac{\partial f_n}{\partial x_n} \end{bmatrix}$$

or, where analytical derivatives are difficult:

$$\frac{\partial f_1}{\partial x_1}(\mathbf{x}^i) \cong \frac{f_1(x_1^i + \delta, x_2^i, \dots, x_n^i) - f_1(x_1^i - \delta, x_2^i, \dots, x_n^i)}{2 \cdot \delta}$$

and so forth.

# Solving Sets of Nonlinear Algebraic Equations

A Python function for Newton's method: **multinewt**

```
def multinewt(f,J,x0,Ea=1.e-7,maxiter=30,decel=1.0):
    xold = x0
    for iter in range(maxiter):
        Jinv = np.linalg.inv(J(xold))
        xnew = xold - decel*Jinv.dot(f(xold))
        xdev = xnew - xold
        xerr = xdev.dot(xdev)
        if xerr < Ea:
            break
        xold = xnew
    return xnew,iter+1
```

**f** : function to evaluate equations' errors

**J**: function to provide the Jacobian

**Ea**: relative error criterion for convergence

**maxiter**: maximum number of iterations

**decel**: decelerator, if needed for stability

- **xnew** computed with the Newton formula using **dot** method
- **xdev** is a vector of x differences from last iteration to this one
- **xerr** is sum of squares of **xdev** elements (inner product with **dot**)
- function returns the latest **xnew** and number of iterations

# Solving Sets of Nonlinear Algebraic Equations

Solution with function `multinewt`

Example

$$\begin{aligned}x^2 + y^2 - 4 &= 0 \\ x \cdot y - 1 &= 0\end{aligned}$$
$$\mathbf{J} = \begin{bmatrix} 2x & 2y \\ y & x \end{bmatrix}$$

```
def f(x):  
    fn = np.zeros(2)  
    fn[0] = x[0]**2 + x[1]**2 - 4  
    fn[1] = x[0]*x[1] - 1  
    return fn
```

```
def J(x):  
    Jac = np.zeros((2,2))  
    Jac[0,0] = 2*x[0]  
    Jac[0,1] = 2*x[1]  
    Jac[1,0] = x[1]  
    Jac[1,1] = x[0]  
    return Jac
```

```
x0 = np.array([0.5, 1.4])  
xsoln, niter = multinewt(f, J, x0)  
print(xsoln)  
print(niter)
```

[0.51763809 1.93185165]

4

**MultiNewton.py**

# Solving Sets of Nonlinear Algebraic Equations

Solution with the **root** function from the SciPy **optimize** submodule

Example  $x^2 + y^2 - 4 = 0$   
 $x \cdot y - 1 = 0$

There is an option to allow the Jacobian matrix to be supplied.

`x = root(f,x0)` without supplying J (Jacobian approximated numerically)

```
import numpy as np
from scipy.optimize import root

def f(x):
    fn = np.zeros(2)
    fn[0] = x[0]**2 + x[1]**2 - 4
    fn[1] = x[0]*x[1]-1
    return fn

x0 = np.array([0.5, 1.4])
result = root(f,x0)

print('\nsolution is',result.x)
print('\n',result)
```

**root** returns a solution object, **result**  
**x** is the solution – **result.x** is the property of **result**

```
solution is [0.51763809 1.93185165]

message: The solution converged.
success: True
status: 1
  fun: [ 4.859e-12  1.982e-11]
   x: [ 5.176e-01  1.932e+00]
 nfev: 11
  fjac: [[-4.791e-01 -8.778e-01]
         [ 8.778e-01 -4.791e-01]]
   r: [-2.129e+00 -1.971e+00  3.232e+00]
  qtf: [ 2.564e-09  6.572e-10]
```

**root\_function\_example.py**

# Solving Sets of Nonlinear Algebraic Equations

Example problem: steam/water equilibrium

$$P \cdot V = \frac{m}{MW} \cdot R \cdot (T + 273.15)$$

ideal gas law

$$\log_{10} P = A - \frac{B}{T + C}$$

Antoine equation

$P$  : absolute pressure, Pa

$A, B, C$  : Antoine constants for H<sub>2</sub>O

$V$  : vapor volume, m<sup>3</sup>

$$A = 11.21 \quad B = 2354.7 \quad C = 280.71$$

$m$  : mass of vapor, kg

$MW$ : H<sub>2</sub>O molecular weight,  $\cong 18.02$  kg/kgmol

$R$  : gas law constant, 8314 (Pa•m<sup>3</sup>)/(kgmol•K)

$T$  : temperature, °C

Operating conditions:  $m = 3.755$  kg       $V = 3.142$  m<sup>3</sup>

Solve for  $P$  and  $T$  .

# Solving Sets of Nonlinear Algebraic Equations

Example problem: steam/water equilibrium

Formulating the problem for solution

$$f_1(T, P) = P \cdot V - \frac{m}{MW} \cdot R \cdot (T + 273.15)$$

$$f_2(T, P) = \log_{10} P - A + \frac{B}{T + C}$$

$$\mathbf{J} \left( \begin{bmatrix} P \\ T \end{bmatrix} \right) = \begin{bmatrix} V & -\frac{m}{MW} \cdot R \\ \frac{1}{\ln(10) \cdot P} & -\frac{B}{(C + T)^2} \end{bmatrix}$$

analytical  
Jacobian  
practical  
in this case

A possible issue here is the comparative scaling of the two equations. Typical values for the PV term could be of magnitude  $10^6$ ; whereas, terms in the second equation are closer to unity. A practical approach to this is to scale the first equation by dividing it by, e.g., 100,000.

$$f_1(T, P) = \left( P \cdot V - \frac{m}{MW} \cdot R \cdot (T + 273.15) \right) / 100000$$

$$f_2(T, P) = \log_{10} P - A + \frac{B}{T + C}$$

$$\mathbf{J} \left( \begin{bmatrix} P \\ T \end{bmatrix} \right) = \begin{bmatrix} V/1e5 & -\frac{m}{MW} \cdot R / 1e5 \\ \frac{1}{\ln(10) \cdot P} & -\frac{B}{(C + T)^2} \end{bmatrix}$$

# Solving Sets of Nonlinear Algebraic Equations

Example problem: steam/water equilibrium

```
import numpy as np
from scipy.optimize import root

R = 8314. # Pa*m3/(kmol*K)
MW = 18.02 # kg/kmol
# Antoine coefficients
A = 11.21 ; B = 2354.7 ; C = 280.7

def SteamEq(x,m,V):
    P = x[0]
    T = x[1]
    ferr = np.zeros(2)
    ferr[0] = (P*V - m/MW*R*(T+273.15))/1e5
    ferr[1] = np.log10(P) - A + B/(T+C)
    return ferr

m = 3.755 # kg
V = 3.142 # m3

x0 = [ 2.e5, 110.]

result = root(SteamEq,x0,args=(m,V))
print('\nsolution is:')
print('P = {0:5.1f} kPa'.format(result.x[0]/1000))
print('T = {0:6.1f} degC'.format(result.x[1]))
```

SteamEquilibrium.py

solution is:  
P = 216.9 kPa  
T = 120.2 degC

# Solving Single Differential Equations

generally, analytical solutions  
are not feasible

Two types

$$\frac{dy}{dt} = f(t) \quad \Longrightarrow \quad \int_{y_0}^{y_f} dy = y_f - y_0 = \underline{\int_{t_0}^{t_f} f(t) dt}$$

finding the area under the curve  
or quadrature

$$\frac{dy}{dt} = f(t, y) \quad \text{numerical methods}$$

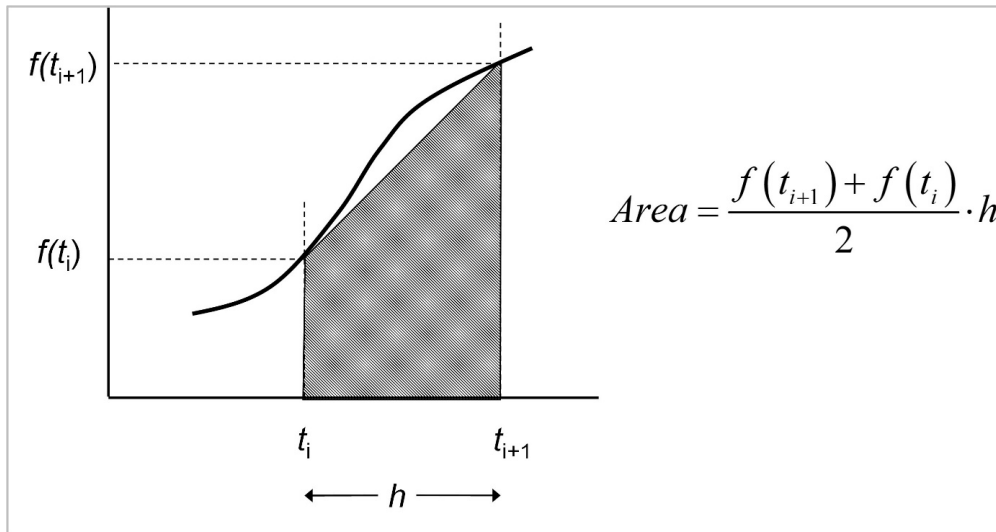
used to solve



# Solving Single Differential Equations

Quadrature – Analytical function

Trapezoidal rule



$$\int_a^b f(t) \cdot dt \cong \left( f(a) + 2f(a + \Delta t) + 2f(a + 2\Delta t) + \cdots + 2f(a + (n-1)\Delta t) + f(b) \right) \frac{h}{2}$$

# Solving Single Differential Equations

Quadrature – Analytical function

$$\frac{dy}{dt} = t \cdot \cos(t) \quad \Rightarrow \quad y = \int_0^{\pi/2} t \cdot \cos(t) \cdot dt$$

Python function for trapezoidal rule, **trap**

```
def trap(f,a,b,n=100):
    x = a # set x to left side a
    h = (b-a)/n # compute interval width
    sm = f(a) # first term of sum
    for i in range(n-1):
        x = x + h # advance x
        sm = sm + 2*f(x) # add 2 * f(x) to sum
    sm = sm + f(b) # add last term to sum
    ar = sm*h/2 # complete integral formula
    return ar

import numpy as np

def f(t):
    return t*np.cos(t)

y = trap(f,0.,np.pi/2)
print(y)
```

0.5707434665276926

# Solving Single Differential Equations

quad\_example.py

Quadrature – Analytical function

– **quad** function from the SciPy **integral** submodule

Standard normal distribution – cumulative probability

$$f(z) = \frac{1}{\sqrt{2\pi}} e^{-\frac{1}{2}z^2} \quad \Rightarrow \quad \frac{dP}{dz} = f(z) \quad \Rightarrow \quad P[-\infty \leq z \leq a] = \int_{-\infty}^a f(z) dz$$

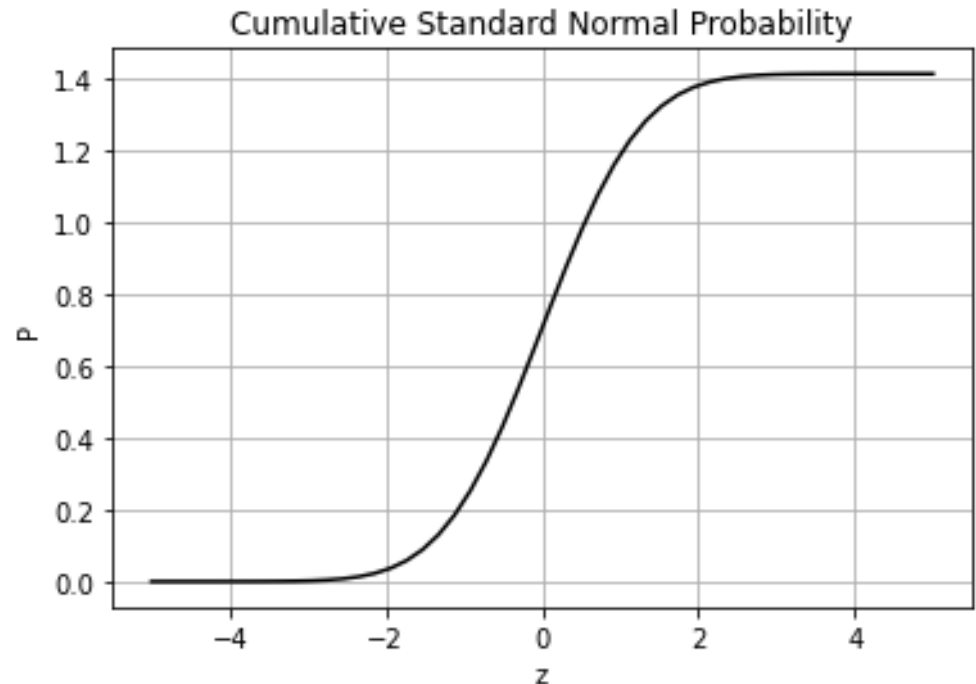
```
import numpy as np
from scipy.integrate import quad
import matplotlib.pyplot as plt

stdnormdens = lambda z: 1/np.sqrt(np.pi)*np.exp(-z**2/2)

a = np.linspace(-5,5)
n = len(a)
P = np.zeros(n)

for i in range(n):
    P[i],ep = quad(stdnormdens,-np.inf,a[i])

plt.plot(a,P,c='k')
plt.grid()
plt.xlabel('z')
plt.ylabel('P')
plt.title('Cumulative Standard Normal Probability')
```



# Solving Single Differential Equations

## Quadrature – Data

## SciPy function `trapz` from the `integrate` submodule

trapz\_example.py

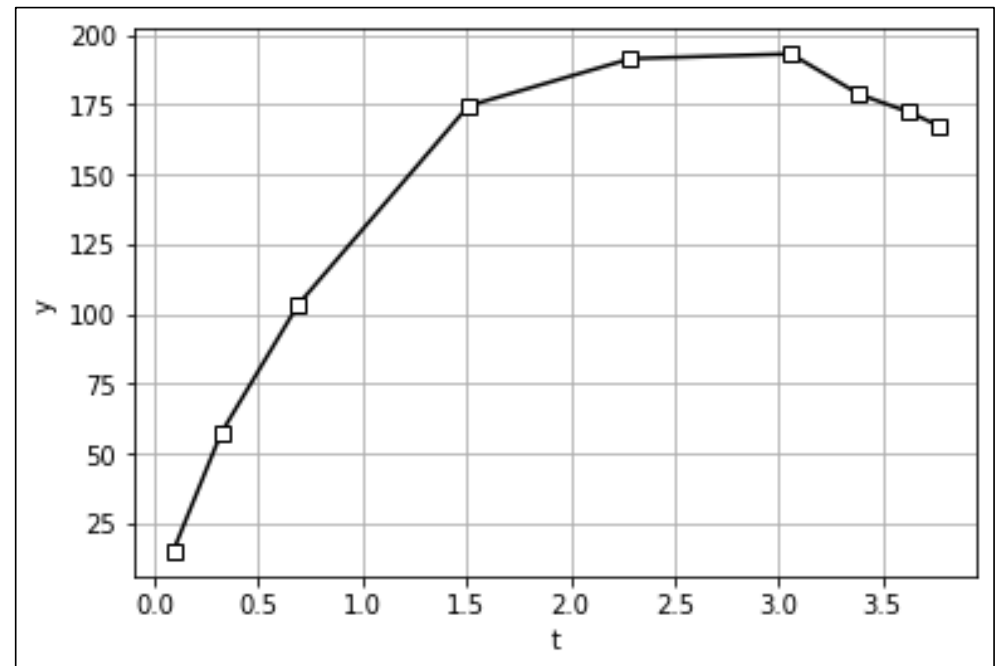
```
import numpy as np
from scipy.integrate import trapz
```

```
t = np.array([0.09,0.32,0.69,1.51,2.29,3.06,3.39,3.63,3.77])
y = np.array([15.1,57.3,103.3,174.6,191.5,193.2,178.7,172.3,167.5])
```

area = 570.1

```
A = trapz(y,t)
print('\narea = {0:5.1f}'.format(A))
```

```
plt.plot(t,y,c='k',marker='s',mec='k',mfc='w')
plt.grid()
plt.xlabel('t')
plt.ylabel('y')
```



# Solving Single Differential Equations

Initial Value Problem

$$\frac{dy}{dt} = f(t, y) \quad y(0) = y_0$$

Example

$$\frac{dy}{dt} = 5(y - t^2) \quad y(0) = 0.08 \quad 0 \leq t \leq 5$$

There is an analytical solution:

$$y = t^2 + 0.4t + 0.08$$

# Solving Single Differential Equations

## Initial Value Problem

### Example using `solve_ivp`

```
import numpy as np
from scipy.integrate import solve_ivp
import matplotlib.pyplot as plt

f = lambda t,y: 5*(y-t**2)

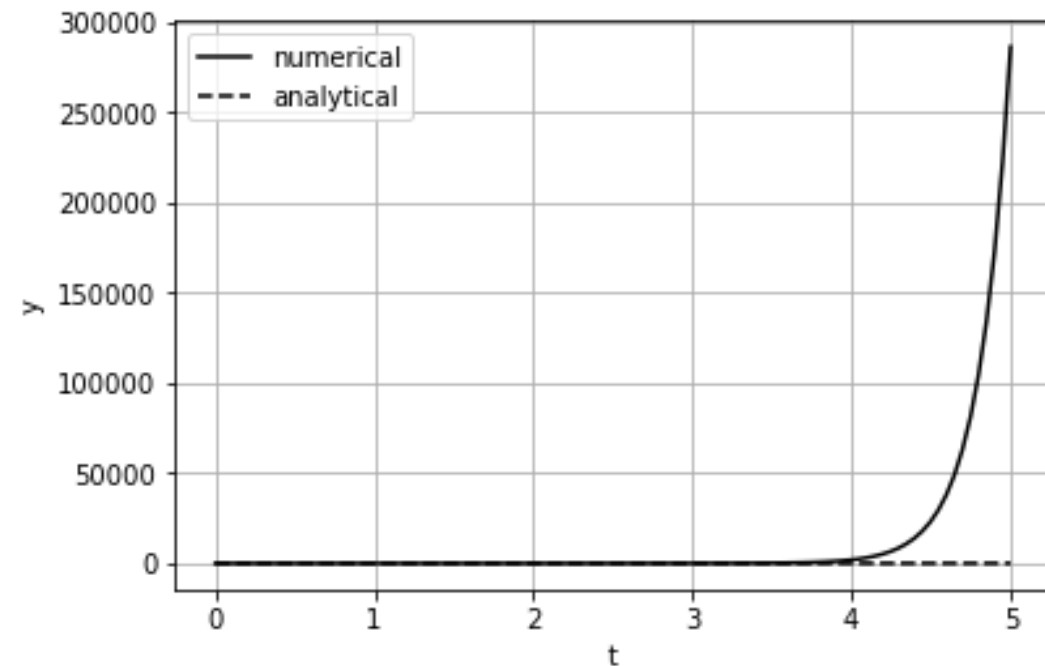
fa = lambda t: t**2 + 0.4*t + 0.08

tspan = [0., 5.]
teval = np.linspace(0.,5.,100)
y0 = [0.08]

result = solve_ivp(f,tspan,y0,t_eval=teval)
tm = result.t
yr = result.y[0,:]

plt.plot(tm,yr,c='k',label='numerical')
plt.plot(tm,fa(tm),c='k',ls='--',label='analytical')
plt.grid()
plt.xlabel('t')
plt.ylabel('y')
plt.legend()
```

Numerical solution blows up!



`parasite.py`

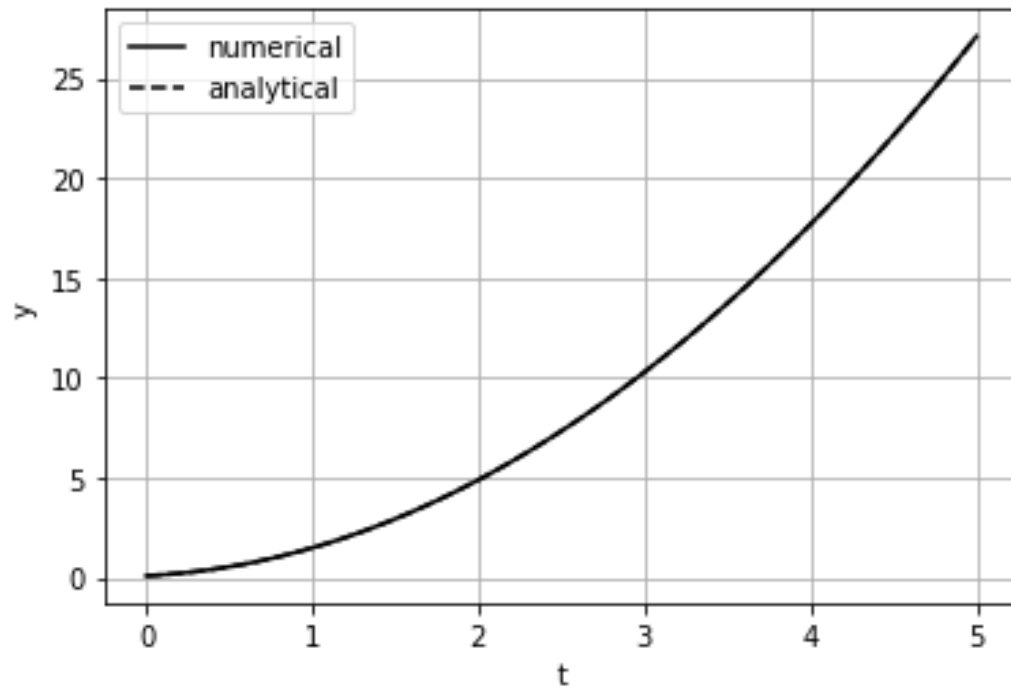
# Solving Single Differential Equations

parasite1.py

Initial Value Problem

Example using `solve_ivp` with tightened tolerances

```
result = solve_ivp(f,tspan,y0,t_eval=teval \
                   , atol=1.e-12, rtol=1.e-12)
```



numerical and analytical  
solutions coincide

# Solving Single Differential Equations

Single Equation Example – Isothermal Batch Reactor  $A + B \xrightarrow{k} C$

Rate of disappearance of A:  $\frac{dC_A}{dt} = -k \cdot C_A \cdot C_B$

Initial conditions:  $C_A(0) = C_{A0}$   $C_B(0) = C_{B0}$   $C_C(0) = C_{C0}$

Basic data:  $k = 14.7 \frac{1}{\text{mol/L} \cdot \text{min}}$

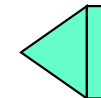
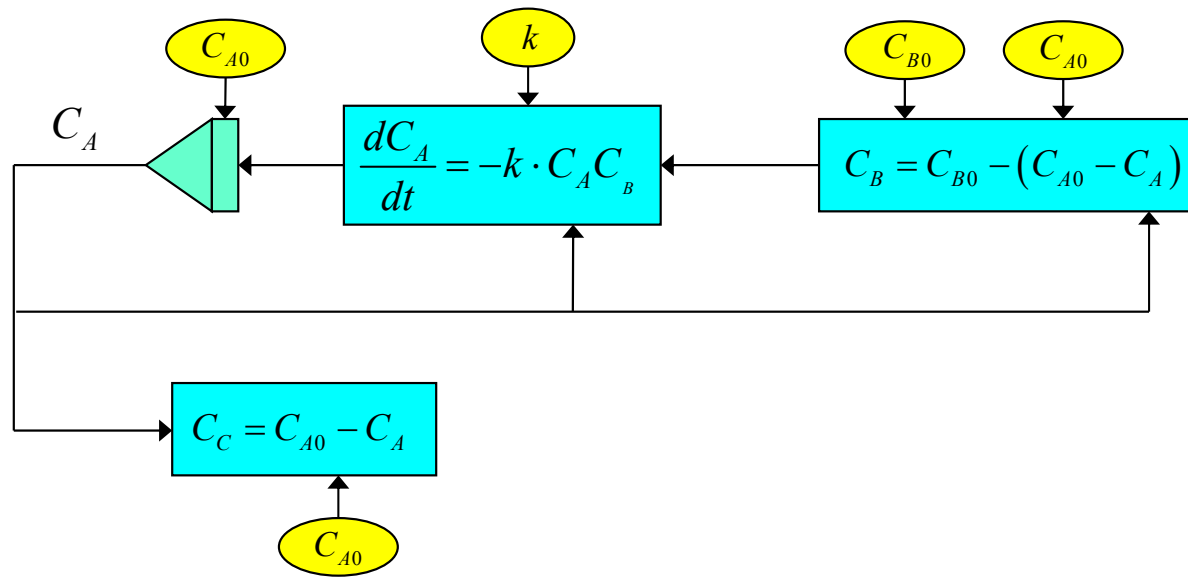
Initial conditions:  $C_{A0} = 0.0209 \frac{\text{mol}}{\text{L}}$   $C_{B0} = C_{A0}/3$   $C_{C0} = 0$

Stoichiometric relationships:  $C_B(t) = C_{B0} - (C_{A0} - C_A(t))$   
 $C_C(t) = C_{C0} + (C_{A0} - C_A(t))$



# Solving Single Differential Equations

Single Equation Example – Isothermal Batch Reactor  
Information Flow Diagram



Integrator

# Solving Single Differential Equations

## Single Equation Example – Isothermal Batch Reactor

```
import numpy as np
from scipy.integrate import solve_ivp
import matplotlib.pyplot as plt

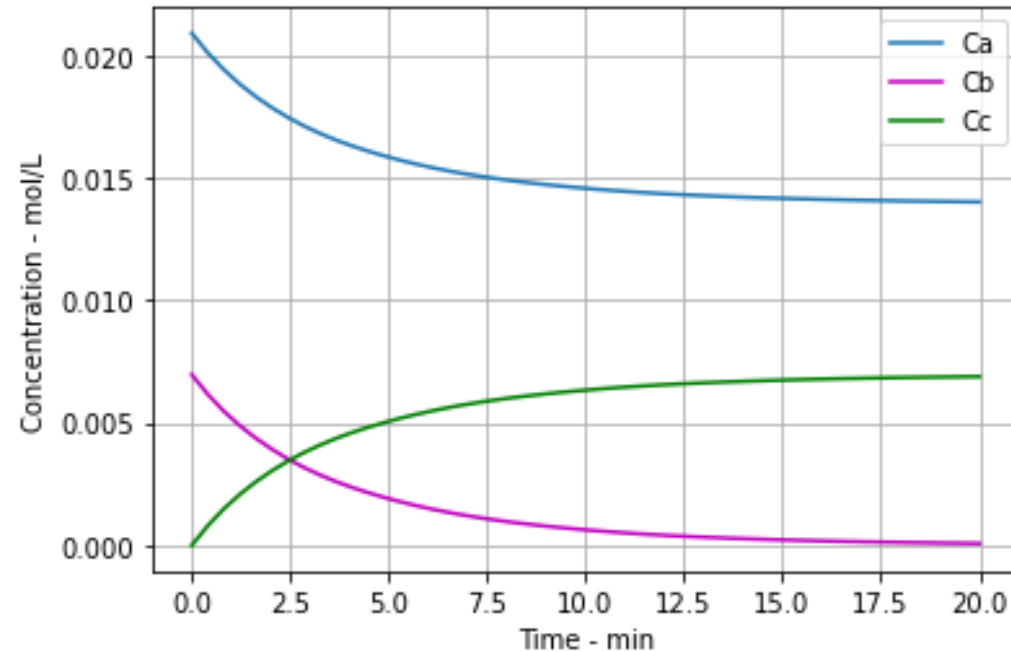
k = 14.7
Ca0 = np.array([0.0209])
Cb0 = Ca0/3

def batch(t, Ca, k, Ca0, Cb0):
    Cb = Cb0 - (Ca0 - Ca)
    return -k*Ca*Cb

tspan = [0., 20.]
teval = np.linspace(0., 20.)
result = solve_ivp(batch, tspan, Ca0, t_eval=teval, args=(k, Ca0, Cb0))

tm = result.t
Ca = result.y[0,:]
Cb = Cb0 - (Ca0 - Ca)
Cc = Ca0 - Ca

plt.plot(tm, Ca, label='Ca')
plt.plot(tm, Cb, c='m', label='Cb')
plt.plot(tm, Cc, c='g', label='Cc')
plt.grid()
plt.xlabel('Time - min')
plt.ylabel('Concentration - mol/L')
plt.legend()
```



SimpleBatch.py

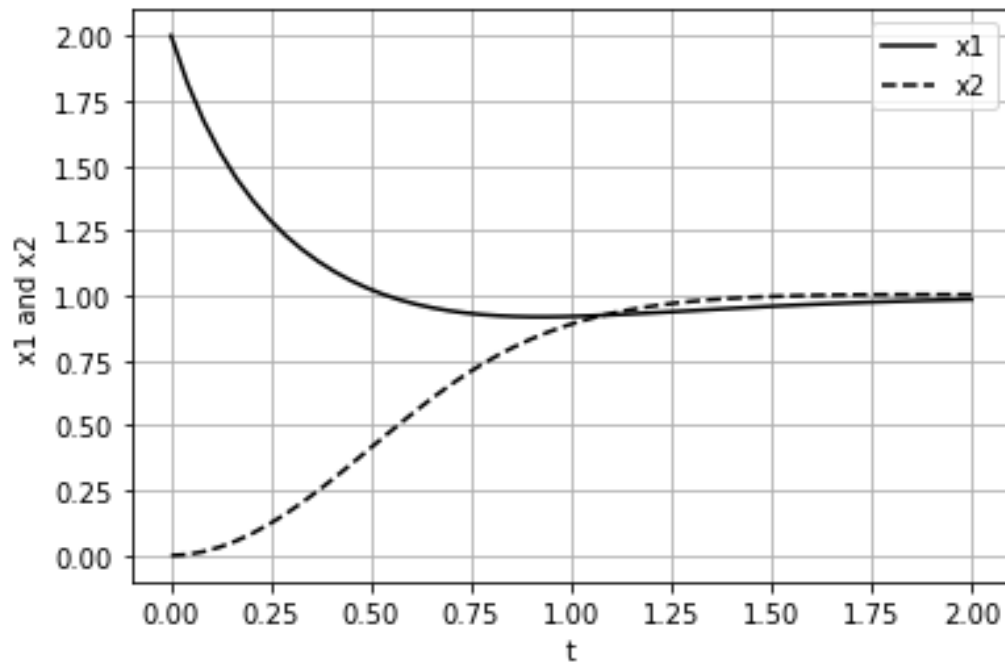
# Solving Multiple Differential Equations

twoODEs.py

Example

$$\frac{dx_1}{dt} = -2x_1^2 + 2x_1 + x_2 - 1 \quad x_1(0) = 2$$

$$\frac{dx_2}{dt} = -x_1 - 3x_2^2 + 2x_2 + 2 \quad x_2(0) = 0$$



```
import numpy as np
from scipy.integrate import solve_ivp
import matplotlib.pyplot as plt

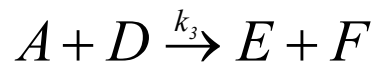
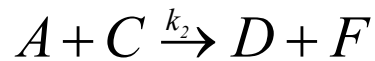
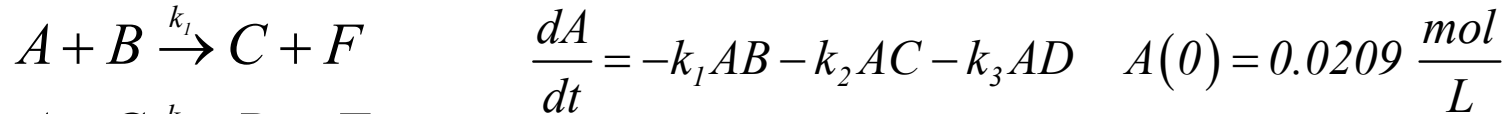
def f(t,y):
    dy = np.zeros(2)
    dy[0] = -2*y[0]**2 + 2*y[0] + y[1] - 1
    dy[1] = -y[0] - 3*y[1]**2 + 2*y[1] + 2
    return dy

tf = 2.
tspan = [0., tf]
teval = np.linspace(0., tf)
y0 = np.array([2., 0.])
result = solve_ivp(f,tspan,y0,t_eval=teval)
tm = result.t
x1 = result.y[0,:]
x2 = result.y[1,:]

plt.plot(tm,x1,c='k',label='x1')
plt.plot(tm,x2,c='k',ls='--',label='x2')
plt.grid()
plt.xlabel('t')
plt.ylabel('x1 and x2')
plt.legend()
```

# Solving Multiple Differential Equations

## Multiple Equation Models – Isothermal Batch Reactor



$$\frac{dB}{dt} = -k_1 AB$$

$$B(0) = \frac{A(0)}{3}$$

$$k_1 = 14.7 \frac{1}{\text{mol/L}} \cdot \frac{1}{\text{min}}$$

$$\frac{dC}{dt} = k_1 AB - k_2 AC$$

$$C(0) = 0$$

$$k_2 = 1.53 \frac{1}{\text{mol/L}} \cdot \frac{1}{\text{min}}$$

$$\frac{dD}{dt} = k_2 AC - k_3 AD$$

$$D(0) = 0$$

$$k_3 = 0.294 \frac{1}{\text{mol/L}} \cdot \frac{1}{\text{min}}$$

From stoichiometry:  $E = \frac{A(0) - A - C - 2D}{3}$  and  $F = A(0) - A$

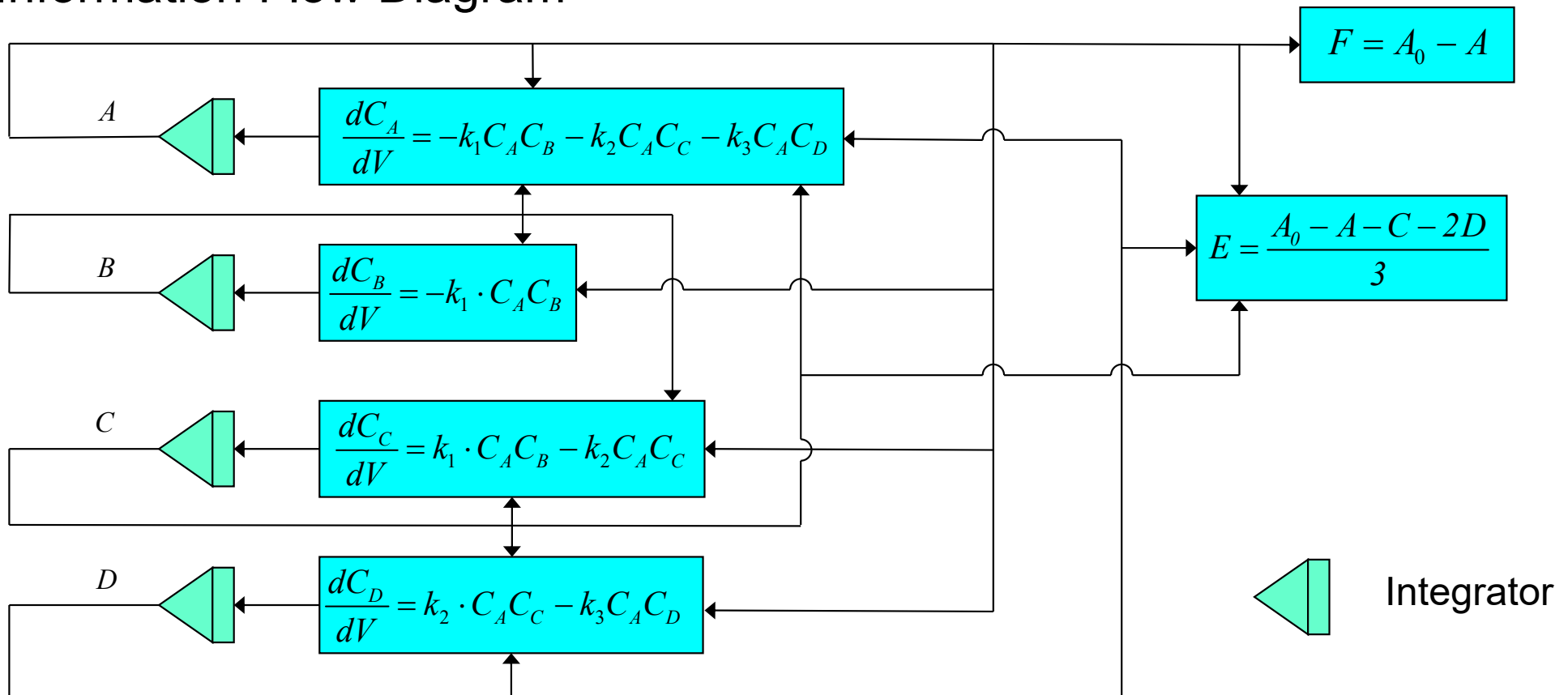
Svirbely, W.J., and J.A. Blauer, *The Kinetics of Three-step Competitive Consecutive Second-order Reactions*, **J. Amer. Chem. Soc.**, **83**, 4115, 1961.

Svirbely, W.J., and J.A. Blauer, *The Kinetics of the Alkaline Hydrolysis of 1,3,5-Tricarbomethoxybenzene*, **J. Amer. Chem. Soc.**, **83**, 4118, 1961.

# Solving Multiple Differential Equations

Multiple Equation Models – Isothermal Batch Reactor

Information Flow Diagram



# Solving Multiple Differential Equations

## Multiple Equation Models – Isothermal Batch Reactor

```
import numpy as np
import matplotlib.pyplot as plt
from scipy.integrate import solve_ivp

def multibatch(t,y,k1,k2,k3):
    A = y[0] ; B = y[1] ; C = y[2] ; D = y[3]
    dA = -k1*A*B - k2*A*C - k3*A*D
    dB = -k1*A*B
    dC = k1*A*B - k2*A*C
    dD = k2*A*C - k3*A*D
    return [dA, dB, dC, dD]
```

additional arguments

unpack y into familiar variables

compute derivatives

# Solving Multiple Differential Equations

## Multiple Equation Models – Isothermal Batch Reactor

```
k1 = 14.7
k2 = 1.53
k3 = 0.294
A0 = 0.0209
B0 = A0/3

tspan = [ 0., 500.]
teval = np.linspace(0.,500.,100)
y0 = [A0, B0, 0, 0]

result = solve_ivp(multibatch,tspan,y0,t_eval=teval,args=(k1,k2,k3))

tm = result.t
A = result.y[0,:]
B = result.y[1,:]
C = result.y[2,:]
D = result.y[3,:]

E = (A0 - A - C - 2*D)/3
F = A0 - A
```

unpack results

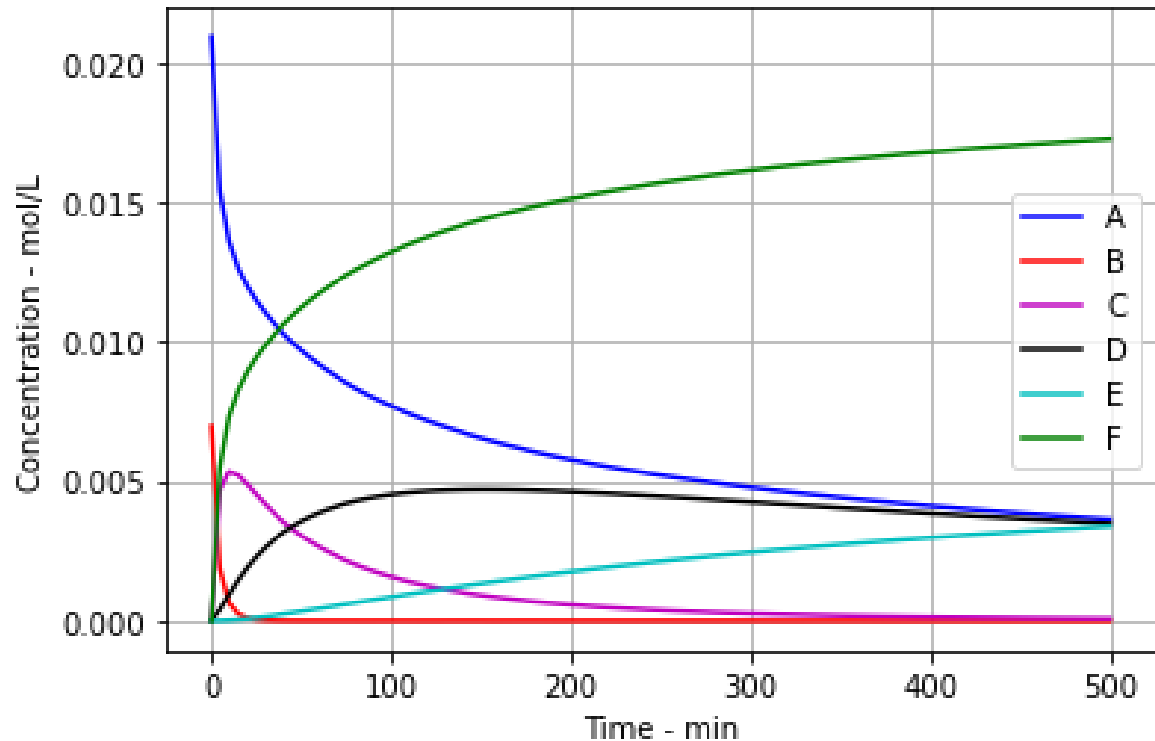
compute E and F  
from A, C and D

# Solving Multiple Differential Equations

## Multiple Equation Models – Isothermal Batch Reactor

```
plt.plot(tm,A,c='b',label='A')
plt.plot(tm,B,c='r',label='B')
plt.plot(tm,C,c='m',label='C')
plt.plot(tm,D,c='k',label='D')
plt.plot(tm,E,c='c',label='E')
plt.plot(tm,F,c='g',label='F')
plt.grid()
plt.xlabel('Time - min')
plt.ylabel('Concentration - mol/L')
plt.legend()
```

**multibatch\_reactor.py**





# Solving Ordinary Differential Equations

Second-order differential equation with split boundary conditions

$$\frac{d^2 y}{dt^2} = \frac{1}{4} \frac{dy}{dt} + y \quad y(0) = 5 \quad y(10) = 8 \quad 0 \leq t \leq 10$$

Decompose into two first-order ODEs

$$\frac{dy}{dt} = y_1 \quad y(0) = 5 \quad y(10) = 8$$

$$\frac{dy_1}{dt} = \frac{1}{4} y_1 + y$$

“Shooting” Strategy

1. Estimate a value for  $y_1$  ( $dy/dt$ ) at  $t = 0$ .
2. Solve the ODEs to  $t = 10$
3. Check  $y(10)$  versus the required value, 8.
4. Adjust the  $y_1(0)$  value and repeat steps 2 and 3 until the desired  $y(10)=8$  value is obtained.

# Solving Ordinary Differential Equations

## Second-order differential equation with split boundary conditions

```
import numpy as np
import matplotlib.pyplot as plt
from scipy.integrate import solve_ivp

def diffeqs(t,y):
    dy1 = y[1]
    dy2 = y[1]/4. + y[0]
    return [ dy1, dy2]

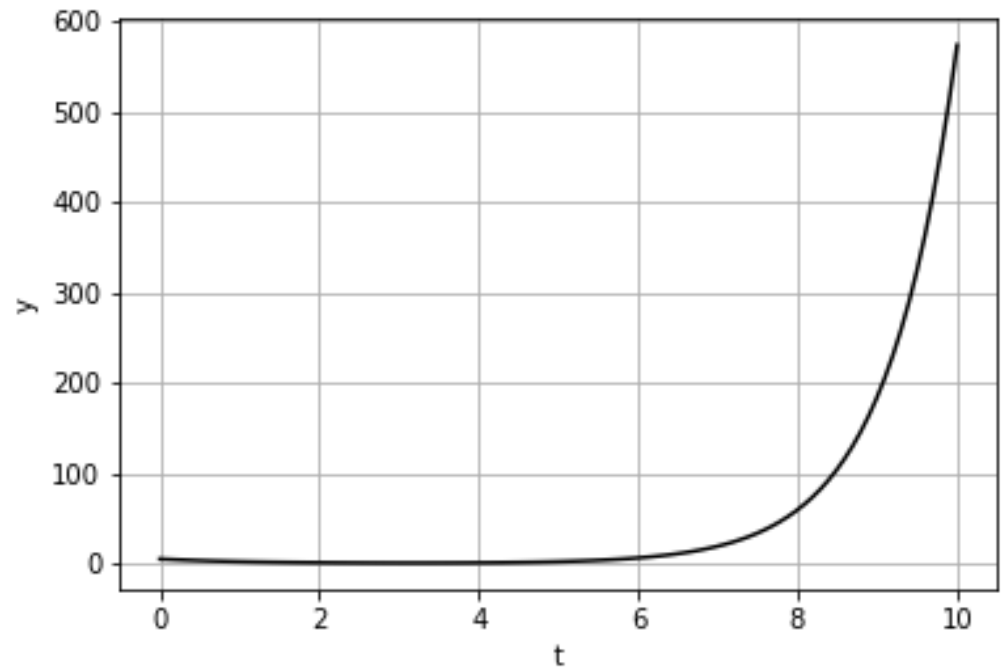
y0 = [5., -4.4]
tspan = [ 0., 10.]
teval = np.linspace(0.,10.,100)

result = solve_ivp(diffeqs,tspan,y0,t_eval=teval)

tm = result.t
y1 = result.y[0,:]
y2 = result.y[1,:]

plt.plot(tm,y1,c='k')
plt.grid()
plt.xlabel('t')
plt.ylabel('y')
```

**TwoPointBC.py**



Equations solved with an estimate for  $y_1(0)$ .  
Clearly doesn't meet the required final condition  $y(10)=8$ .

# Solving Ordinary Differential Equations

Second-order differential equation with split boundary conditions

Employ **brentq** to satisfy the final boundary condition

```
import numpy as np
import matplotlib.pyplot as plt
from scipy.integrate import solve_ivp
from scipy.optimize import brentq

def diffeqs(t,y):
    dy1 = y[1]
    dy2 = y[1]/4. + y[0]
    return [ dy1, dy2]

def splitboundary(y01):
    y0 = [5., y01]
    tspan = [ 0., 10.]
    teval = np.linspace(0.,10.,100)
    result = solve_ivp(diffeqs,tspan,y0,t_eval=teval)
    tm = result.t
    n = len(tm)
    return result.y[0,n-1]-8.

y01 = -4.4
y1_soln = brentq(splitboundary,-5.,-4.)
print('initial derivative value =',y1_soln)

y0 = [5., y1_soln]
tspan = [ 0., 10.]
teval = np.linspace(0.,10.,100)
result = solve_ivp(diffeqs,tspan,y0,t_eval=teval)
tm = result.t
y1 = result.y[0,:]

plt.plot(tm,y1,c='k')
plt.grid()
plt.xlabel('t')
plt.ylabel('y')
```

**TwoPointBC1.py**

# Solving Ordinary Differential Equations

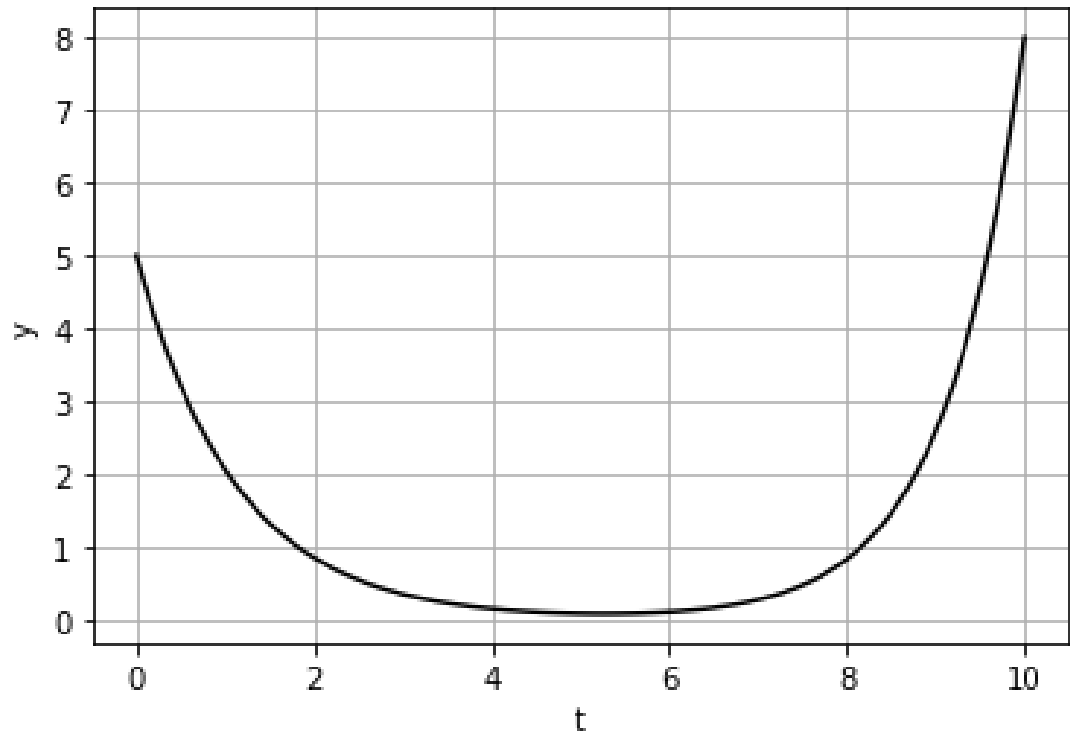
Second-order differential equation with split boundary conditions

Employ **brentq** to satisfy the final boundary condition

Final condition,  $y = 8.$ ,  
is now met.

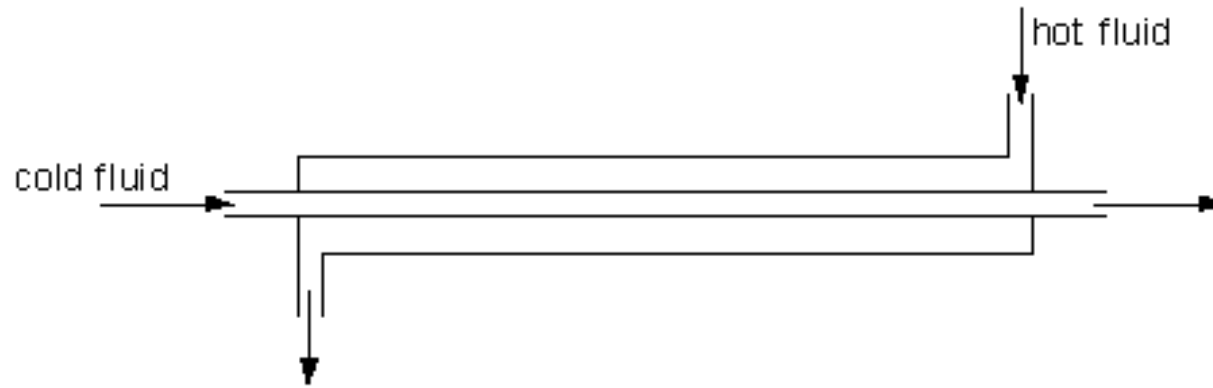
`initial derivative value = -4.413717066724333`

Final condition is very sensitive  
to the initial derivative value.



# Solving Ordinary Differential Equations

Example: tube-in-tube, countercurrent heat exchanger



$$\frac{dT_c}{dz} = \frac{h_i A_i}{w_c C_c} (T_h - T_c)$$

$$T_c(0) = T_{ci}$$

$$\frac{dT_h}{dz} = \frac{h_o A_o}{w_h C_h} (T_h - T_c)$$

$$T_h(L) = T_{hi}$$

$$h_o = \frac{h_i \cdot D_i}{D_o}$$

# Solving Ordinary Differential Equations

## Example: tube-in-tube, countercurrent heat exchanger

$z$ : distance down the heat exchanger from the cold fluid inlet (on the left)

$L$ : length of the heat exchanger

$T_c$ : temperature of the cold fluid, a function of  $z$

$T_{ci}$ : cold water inlet temperature, at  $z=0$

$T_{hi}$ : hot water inlet temperature, at  $z=L$

$T_h$ : temperature of the hot fluid, a function of  $z$

$w_c$ : mass flow rate of cold fluid

$w_h$ : mass flow rate of hot fluid

$C_c$ : heat capacity of cold fluid

$C_h$ : heat capacity of hot fluid

$A_i$ : inside area for heat transfer (cold fluid) per unit length

$A_o$ : outside area for heat transfer (hot fluid) per unit length

$h_i$ : inside heat transfer coefficient (cold fluid)

$h_o$ : outside heat transfer coefficient (hot fluid)

# Solving Ordinary Differential Equations

Example: tube-in-tube, countercurrent heat exchanger

$$\frac{dT_c}{dz} = \frac{h_i A_i}{w_c C_c} (T_h - T_c) \quad T_c(0) = T_{ci}$$

$$\frac{dT_h}{dz} = \frac{h_o A_o}{w_h C_h} (T_h - T_c) \quad T_h(L) = T_{hi}$$

The issue we have with solving these equations is that the cold stream boundary condition is at  $z = 0$  and the hot stream boundary condition is at  $z = L$ , the other end of the heat exchanger. A practical way to handle this is to estimate the hot stream temperature at  $z = 0$ , proceed with the solution, and adjust that estimate later on to meet the condition at  $z = L$ .

# Solving Ordinary Differential Equations

Example: tube-in-tube, countercurrent heat exchanger

Basic data and operating conditions

Outer tube	Inner tube	Length 5 m
11 BWG	11 BWG	
OD 2 in, ID 1.76 in	OD 1 in, ID 0.76 in	

Inlet temperatures	Fluid density (H <sub>2</sub> O)	Heat capacity (H <sub>2</sub> O)
Hot stream 50 °C	988 kg/m <sup>3</sup>	4187 J/(kg·°C)
Cold stream 10 °C		

Hot stream flow rate	1 L/s	Heat transfer coefficient
Cold stream	0.3 L/s	$h_i = 14,000 \text{ W}/(\text{m}^2 \cdot ^\circ\text{C})$



# Solving Ordinary Differential Equations

Example: tube-in-tube, countercurrent heat exchanger

```
import numpy as np
import matplotlib.pyplot as plt
from scipy.integrate import solve_ivp

def htxr(z, T, hi, ho, Ai, Ao, wc, wh, cP):
    Tc = T[0]
    Th = T[1]
    dTc = hi*Ai/wc/cP*(Th-Tc)
    dTh = ho*Ao/wh/cP*(Th-Tc)
    return [dTc, dTh]

doin = 1. # in
do = doin * 0.0254 # m
Ao = np.pi*do # m2/m
diin = 0.76 # in
di = diin * 0.0254 # m
Ai = np.pi*di # m2/m
L = 5. # m

den = 998. # kg/m3
cP = 4187. # J/(kg*degC)

hi = 14000. # W/(m2*degC)
ho = hi*di/do
```

```
qcL = 0.3 # L/s
qc = qcL/1000. # m3/s
wc = qc * den # kg/s
qhL = 1. # L/s
qh = qhL/1000. # m3/s
wh = qh * den # kg/s

Tci = 10 # degC
Thi = 50 # degC

zspan = [0., L]
zeval = np.linspace(0., L, 100)

Tho = 40 # degC ←
T0 = [ Tci, Tho]

result = solve_ivp(htxr, zspan, T0, t_eval=zeval \
, args=(hi, ho, Ai, Ao, wc, wh, cP))

zs = result.t
Tc = result.y[0, :]
Th = result.y[1, :]

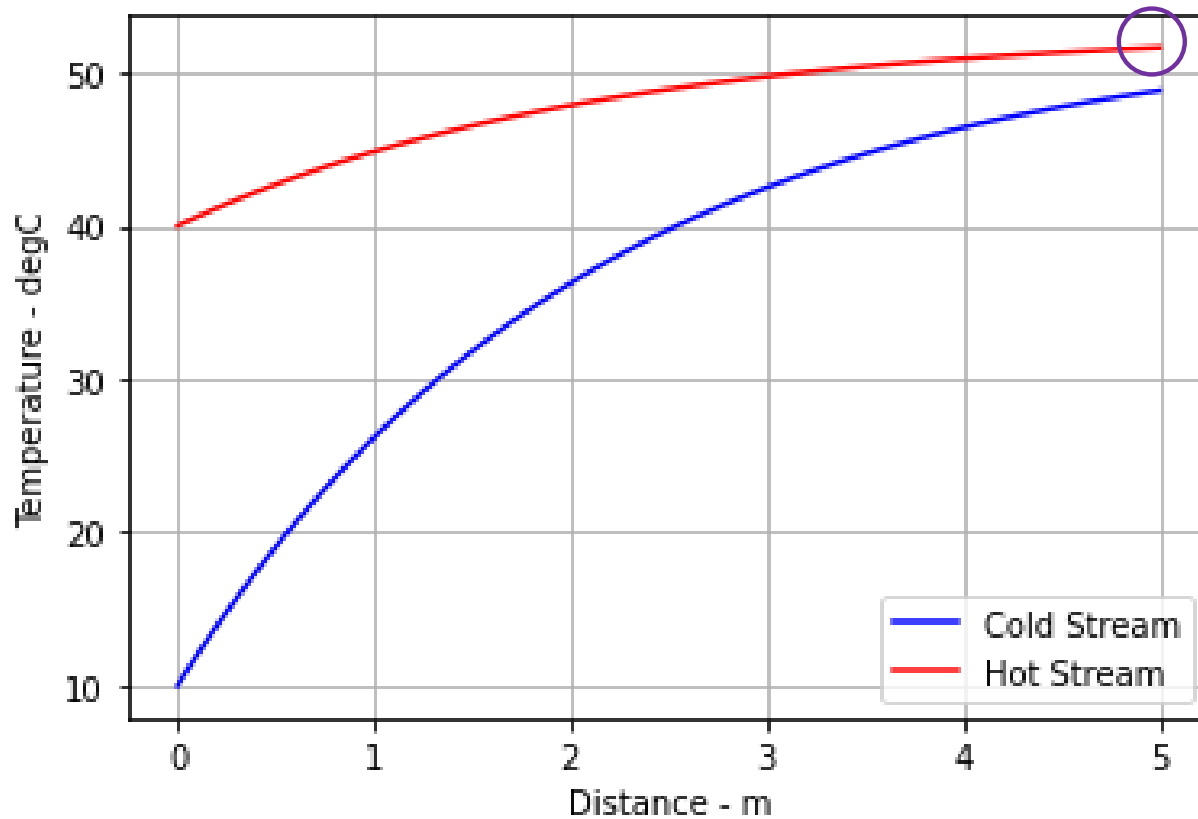
plt.plot(zs, Tc, c='b', label='Cold Stream')
plt.plot(zs, Th, c='r', label='Hot Stream')
plt.grid()
plt.xlabel('Distance - m')
plt.ylabel('Temperature - degC')
plt.legend()
```

countercurrent\_heatexchanger1.py

Solve the model first with  
an estimate for the hot stream  
outlet temperature.

# Solving Ordinary Differential Equations

Example: tube-in-tube, countercurrent heat exchanger



Hot stream inlet condition, 50°C, not met.

# Solving Ordinary Differential Equations

Example: tube-in-tube, countercurrent heat exchanger

Use the **brentq** function to adjust the hot stream outlet temperature until the inlet hot stream condition is met.

```
import numpy as np
import matplotlib.pyplot as plt
from scipy.integrate import solve_ivp
from scipy.optimize import brentq

def htxr(z,T,hi,ho,Ai,Ao,wc,wh,cP):
    Tc = T[0]
    Th = T[1]
    dTc = hi*Ai/wc/cP*(Th-Tc)
    dTh = ho*Ao/wh/cP*(Th-Tc)
    return [dTc, dTh]

def findTho(Tho,Thi_spec):
    zspan = [0., L]
    T0 = [ Tci, Tho]
    result = solve_ivp(htxr,zspan,T0 \
                      ,args=(hi,ho,Ai,Ao,wc,wh,cP))
    zs = result.t
    n = len(zs)
    return result.y[1,n-1] - Thi_spec
```

**countercurrent\_heatexchanger2.py**

# Solving Ordinary Differential Equations

Example: tube-in-tube, countercurrent heat exchanger

```
doin = 1. # in
do = doin * 0.0254 # m
Ao = np.pi*do # m2/m
diin = 0.76 # in
di = diin * 0.0254 # m
Ai = np.pi*di # m2/m
L = 5. # m

den = 998. # kg/m3
cP = 4187. # J/(kg*degC)

hi = 14000. # W/(m2*degC)
ho = hi*di/do

qcL = 0.3 # L/s
qc = qcL/1000. # m3/s
wc = qc * den # kg/s
qhL = 1. # L/s
qh = qhL/1000. # m3/s
wh = qh * den # kg/s
```

# Solving Ordinary Differential Equations

Example: tube-in-tube, countercurrent heat exchanger

```
Tci = 10 # degC
Thi_spec = 50 # degC
Tho_soln = brentq(findTho,35.,45.,args=(Thi_spec))

zspan = [0., L]
zeval = np.linspace(0.,L,100)
T0 = [Tci, Tho_soln]
result = solve_ivp(htxr,zspan,T0,t_eval=zeval \
                  ,args=(hi,ho,Ai,Ao,wc,wh,cP))

zs = result.t
Tc = result.y[0,:]
Th = result.y[1,:]

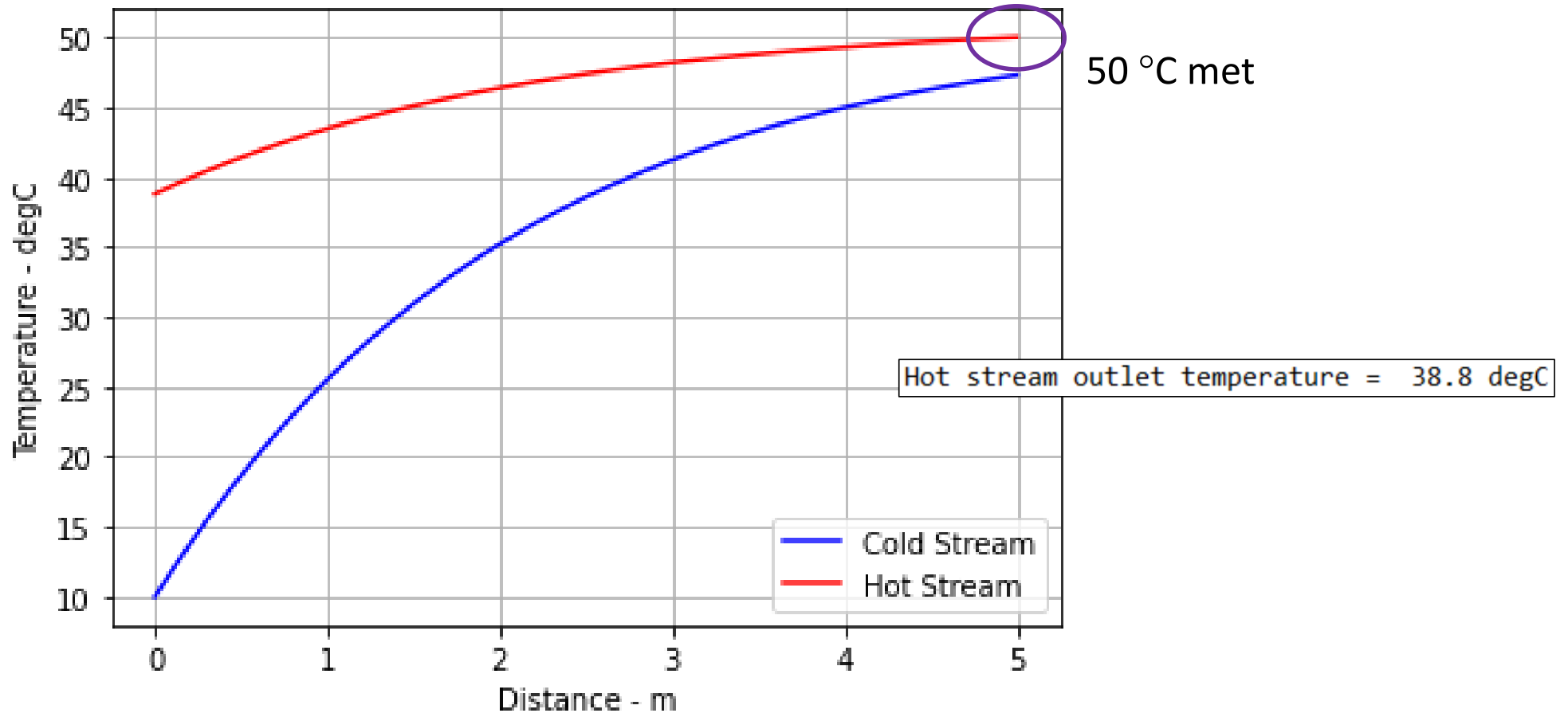
print('Hot stream outlet temperature = {0:5.1f} degC'.format(Tho_soln))

plt.plot(zs,Tc,c='b',label='Cold Stream')
plt.plot(zs,Th,c='r',label='Hot Stream')
plt.grid()
plt.xlabel('Distance - m')
plt.ylabel('Temperature - degC')
plt.legend()
```

**brentq** adjusts **Tho** until **Thi** computed meets the spec, **Thi\_spec**

# Solving Ordinary Differential Equations

Example: tube-in-tube, countercurrent heat exchanger



# Optimization

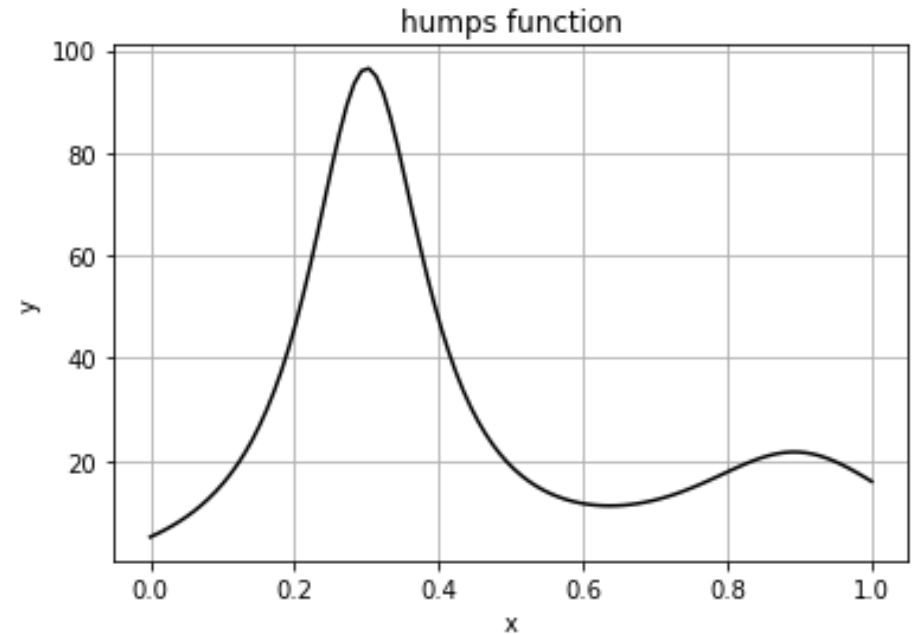
Finding a maximum or minimum of a function with a single adjustable variable

Example 
$$y = \frac{1}{(x-0.3)^2 + 0.01} + \frac{1}{(x-0.9)^2 + 0.04} - 6$$

```
import numpy as np
import matplotlib.pyplot as plt

def humps(x):
    y = 1./((x-0.3)**2+0.01) + 1./((x-0.9)**2+0.04) - 6.
    return y

x = np.linspace(0.,1.,100)
plt.plot(x,humps(x),c='k')
plt.grid()
plt.xlabel('x')
plt.ylabel('y')
plt.title('humps function')
```



**plothumps.py**

# Optimization

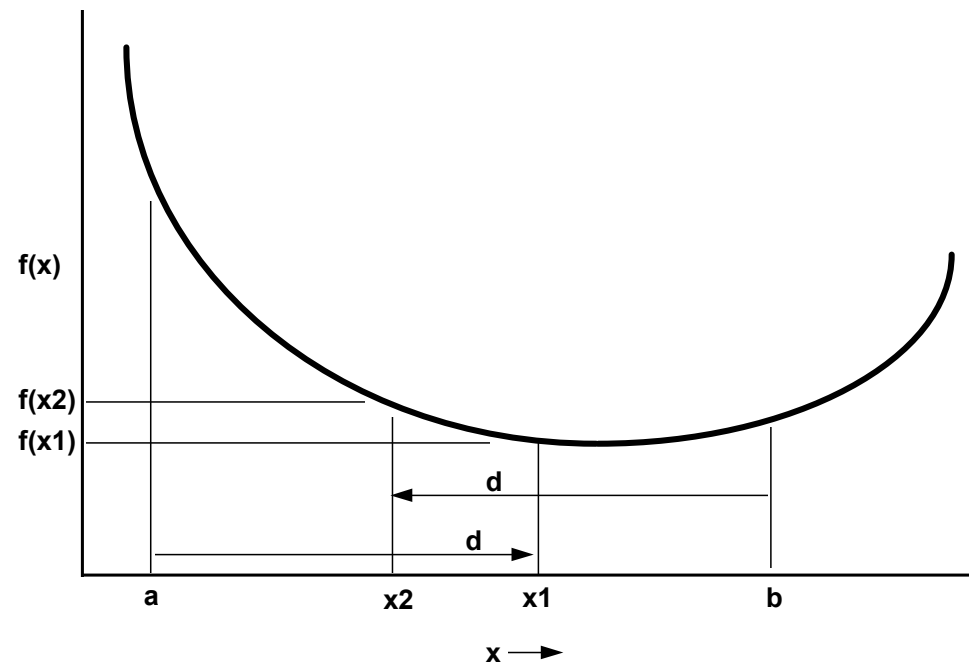
Finding a maximum or minimum of a function with a single adjustable variable - the Golden Section search

This is a bracketing method, similar to bisection. The figure shows a curve  $f(x)$  with a minimum between two initial estimates,  $a$  and  $b$ . Instead of using the midpoint between  $a$  and  $b$ , an overlapping interval  $d$  is used to compute  $x_1$  and  $x_2$ . The interval is given by

$$d = \frac{\sqrt{5}-1}{2} \cdot (b-a)$$

where  $\frac{\sqrt{5}-1}{2}$  is the Golden Ratio ( $GR$ )

with the unique property  $GR = \frac{1}{1+GR}$





# Optimization

Finding a maximum or minimum of a function with a single adjustable variable - the Golden Section search

For the figure to the right, we can see that

$$f(x_2) > f(x_1)$$

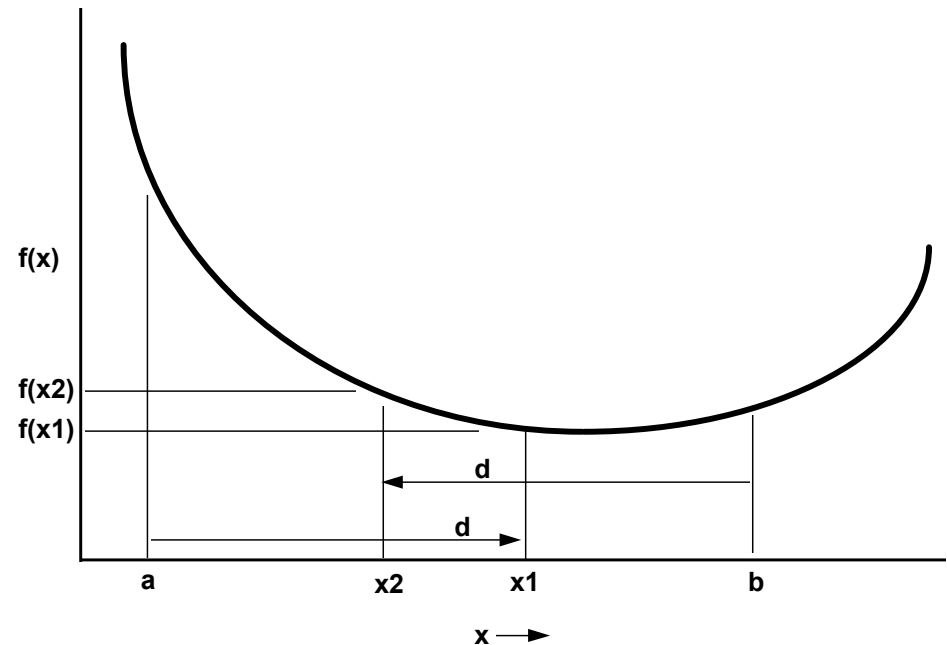
and that leads to the conclusion that the minimum must be between  $x_2$  and  $b$ , and the interval  $[a, x_2]$  can be excluded. This implies that  $x_2$  becomes the new  $a$  for the next iteration of the method.

If

$$f(x_1) > f(x_2)$$

the interval  $[x_1, b]$  would be excluded and  $x_1$  would become the next  $b$ .

For each iteration, the interval containing the minimum is reduced by a factor of  $GR$ .



Any ratio other than  $GR$  that is greater than 0.5 and less than 1 could be used. The advantage of using the  $GR$  is that, for the first scenario above,  $x_1$  is the  $x_2$  value for the next iteration, avoiding the need to compute  $f(x_2)$  then.

# Optimization

Python function to implement Golden Section search:

goldmin\_1.py

```
def mingold(f,a,b,maxit=30):
    GR = (np.sqrt(5)-1)/2
    d = GR*(b-a)
    x1 = a + d
    x2 = b - d
    f1 = f(x1)
    f2 = f(x2)
    for i in range(maxit):
        d = GR*d
        if f2 > f1:
            a = x2
            x2 = x1
            f2 = f1
            x1 = a + d
            f1 = f(x1)
        else:
            b = x1
            x1 = x2
            f1 = f2
            x2 = b - d
            f2 = f(x2)
    return (x1+x2)/2
```

Additional script to solve for minimum of humps function:

```
def humps(x):
    hmps = 1/((x-0.3)**2+0.01)+1/((x-0.9)**2+0.04)-6
    return hmps

a = 0.5
b = 0.8
xmin = mingold(humps,a,b)
print('minimum x = {0:6.4f}'.format(xmin))
```

minimum x = 0.6370

Because of the overlap coincidence, only one function evaluation is needed for each iteration.

The  $d$  value is reduced by  $GR$  each iteration. For 30 iterations, the original  $b - a$  is reduced by  $GR^{30} \cong 5 \times 10^{-7}$ .

# Optimization

Finding the minimum of a function with a single adjustable variable  
Using the **minimize\_scalar** function from the SciPy **optimize** submodule

```
from scipy.optimize import minimize_scalar

def humps(x):
    y = 1./((x-0.3)**2+0.01) + 1./((x-0.9)**2+0.04) - 6.
    return y

result = minimize_scalar(humps)
print(result)
```

```
message:
    Optimization terminated successfully;
    The returned value satisfies the termination criteria
    (using xtol = 1.48e-08 )
success: True
    fun: -6.0
    x: -111097479.90657699
    nit: 35
    nfev: 110
```

determines a minimum well out of range  
expected for x

```
from scipy.optimize import minimize_scalar

def humps(x):
    y = 1./((x-0.3)**2+0.01) + 1./((x-0.9)**2+0.04) - 6.
    return y

result = minimize_scalar(humps, bounds=(0.4,0.8))
print(result)
```

```
message: Solution found.
success: True
status: 0
    fun: 11.252754126374835
    x: 0.6370104788470293
    nit: 9
    nfev: 9
```

the local  
minimum  
is determined

the interval is bounded

**minimize\_humps.py**

# Optimization

Finding the maximum of a function with a single adjustable variable

Negate the function to determine a maximum

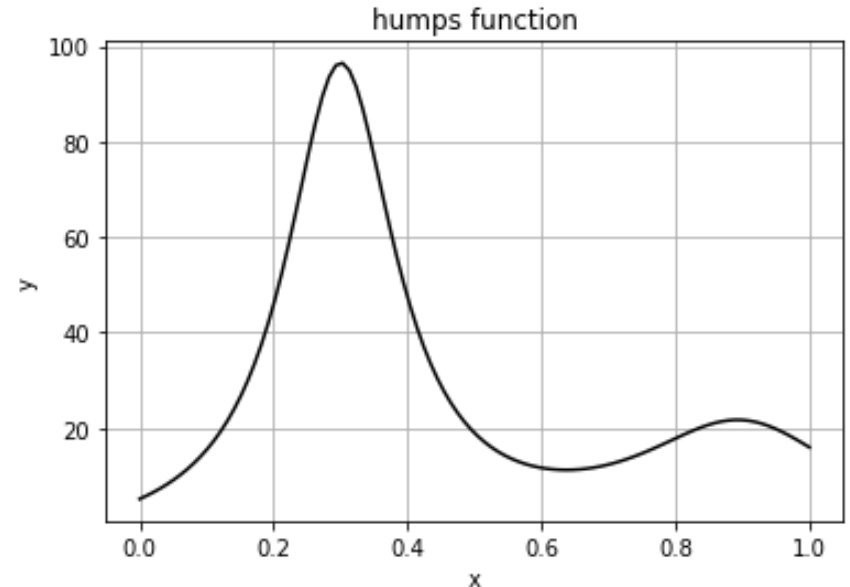
`maximize_humps.py`

```
from scipy.optimize import minimize_scalar

def humps(x):
    y = 1./((x-0.3)**2+0.01) + 1./((x-0.9)**2+0.04) - 6.
    return -y

result = minimize_scalar(humps,bounds=(0.0,1.0))
print(result)
```

```
message: Solution found.
success: True
status: 0
  fun: -96.50140855598092
   x: 0.30037628482912704
  nit: 11
 nfev: 11
```



Could restrict bounds to 0.8 – 1.0 to find local maximum there.



# Optimization

Finding a maximum or minimum of a function with multiple adjustable variables and one or more constraints

Example:

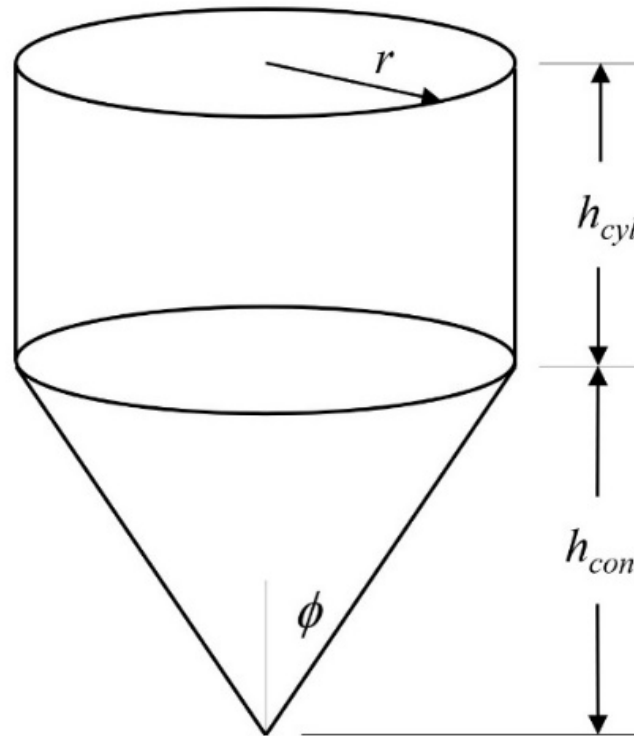
Optimal grain bin design

Minimize surface area  
not including the top

Constraints

$$V = V_{cyl} + V_{con} = 10 \text{ m}^3$$

$$\phi_{\max} = 20.4^\circ$$



$$V_{cyl} = \pi r^2 h_{cyl}$$

$$S_{cyl} = 2\pi r h_{cyl}$$

$$V_{con} = \frac{1}{3} \pi r^2 h_{con}$$

$$S_{con} = \pi r \sqrt{r^2 + h_{con}^2}$$

$$\phi = \tan^{-1} \left( \frac{r}{h_{con}} \right)$$

# Optimization

Example: Optimal grain bin design

Function definitions

grainbin.py

```
import numpy as np
from scipy.optimize import minimize
```

```
def S(x): # bin surface area
    r = x[0]
    hcyl = x[1]
    hcon = x[2]
    Scyl = 2*np.pi*r*hcyl
    Scon = np.pi*r*np.sqrt(r**2+hcon**2)
    return Scyl+Scon
```

```
def V(x): # bin volume
    r = x[0]
    hcyl = x[1]
    hcon = x[2]
    Vcyl = np.pi*r**2*hcyl
    Vcon = np.pi*r**2*hcon/3
    return Vcyl+Vcon
```

```
def VolCon(x): # volume equality constraint
    return V(x)-Volspec
```

```
def AngCon(x): # cone angle inequality constraint
    r = x[0]
    hcon = x[2]
    Angrad = np.arctan(r/hcon)
    Angdeg = np.rad2deg(Angrad)
    return Angspec - Angdeg # must be >= 0
```

# Optimization

Example: Optimal grain bin design

Main script

```
Volspec = 10 # m
Angspec = 20.4 # deg
x0 = [1., 1., 1.] # m
con = ({'type':'eq','fun':VolCon},
      {'type':'ineq','fun':AngCon})

result = minimize(S,x0,constraints=con)

r = result.x[0]
hcyl = result.x[1]
hcon = result.x[2]
print('radius = {0:5.2f} m'.format(r))
print('cylinder height = {0:5.2f} m'.format(hcyl))
print('cone height = {0:5.2f} m'.format(hcon))
phi = np.arctan(r/hcon)
phid = np.rad2deg(phi)
print('cone angle = {0:5.1f} deg'.format(phid))
print('total surface area = {0:6.1f} m2'.format(S(result.x)))
print('bin volume = {0:5.1f} m3'.format(V(result.x)))
```

## Results

```
radius = 1.44 m
cylinder height = 0.26 m
cone height = 3.86 m
cone angle = 20.4 deg
total surface area = 20.9 m2
bin volume = 10.0 m3
```

Most of the bin is the  
conical part

Angle is at the constraint  
Volume constraint met



# Curve-Fitting

Polynomial regression

$\text{coeff} = \text{polyfit}(x, y, \text{order})$

$y = \text{polyval}(\text{coeff}, x)$

Example:

Density of Methanol-  
Water Solutions at  
20 degC

Wt% Methanol	Density (kg/m <sup>3</sup> )
0	998.2
10	981.5
20	966.6
30	951.5
40	934.5
50	915.6
60	894.6
70	871.5
80	846.9
90	820.2
100	791.7

## polynomialfit.py

```
import numpy as np
import matplotlib.pyplot as plt

pct = np.arange(0,110,10)
Den = np.array([998.2, 981.5, 966.6, 951.5, 934.5,
               915.6, 894.6, 871.5, 846.9, 820.2, 791.7])

coeff = np.polyfit(pct,Den,6)
print(coeff)

pctp = np.linspace(0,100,100)
DenP = np.polyval(coeff,pctp)

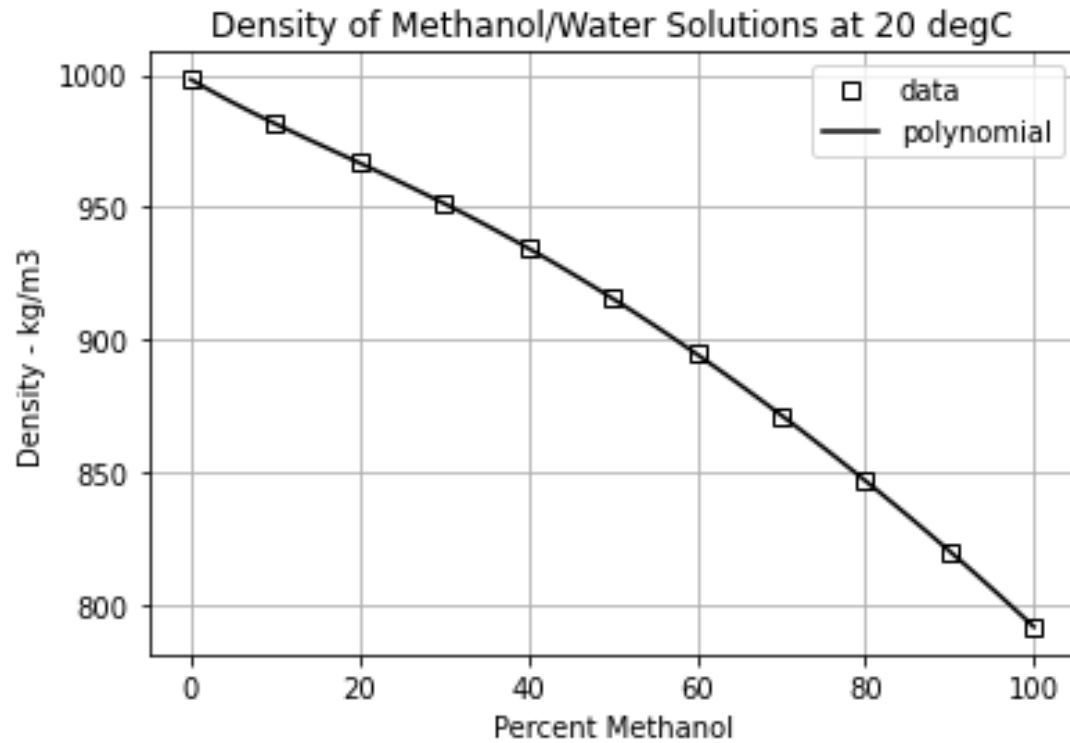
plt.plot(pct,Den,c='w',marker='s',mec='k',mfc='w',label='data')
plt.plot(pctp,DenP,c='k',label='polynomial')
plt.grid()
plt.xlabel('Percent Methanol')
plt.ylabel('Density - kg/m3')
plt.legend()
plt.title('Density of Methanol/Water Solutions at 20 degC')
```

# Curve-Fitting

Polynomial regression

Coefficients from high to low order

```
[ 1.07843137e-10 -5.09426848e-08 9.31033183e-06 -8.37157206e-04  
2.90083656e-02 -1.88888510e+00 9.98205759e+02]
```



# Curve-Fitting

## Multilinear regression

**Model** 
$$y = \beta_0 + \beta_1 f_1(x_j, j = 1, \dots, m) + \beta_2 f_2(x_j, j = 1, \dots, m) + \dots + \beta_k f_k(x_j, j = 1, \dots, m)$$

measurement responses

$$\mathbf{y} = \begin{bmatrix} y_1 \\ y_2 \\ \vdots \\ y_n \end{bmatrix}$$

realization matrix of input levels

$$\mathbf{X} = \begin{bmatrix} 1 & f_{11} & f_{21} & \cdots & f_{m1} \\ 1 & f_{12} & f_{22} & \cdots & f_{m2} \\ \vdots & \vdots & \vdots & \ddots & \vdots \\ 1 & f_{1n} & f_{2n} & \cdots & f_{mn} \end{bmatrix}$$

intercept

parameter estimates

$$\hat{\boldsymbol{\beta}} = \begin{bmatrix} \hat{\beta}_1 \\ \hat{\beta}_2 \\ \vdots \\ \hat{\beta}_k \end{bmatrix}$$

**Dataset**

$$\{y_i, x_{1i}, \dots, x_{mi}, i = 1, \dots, n\}$$

Minimize  $V$  by choice of  $\hat{\boldsymbol{\beta}}$  via calculus

Normal equations

$$(\mathbf{X}^T \mathbf{X}) \mathbf{b} = \mathbf{X}^T \mathbf{y}$$

Fitted model parameters

$$\mathbf{b} = (\mathbf{X}^T \mathbf{X})^{-1} \mathbf{X}^T \mathbf{y}$$

model predictions

$$\hat{\mathbf{y}} = \mathbf{X} \hat{\boldsymbol{\beta}}$$

residuals

$$\mathbf{e} = \begin{bmatrix} e_1 \\ e_2 \\ \vdots \\ e_n \end{bmatrix} \quad \mathbf{e} = \mathbf{y} - \hat{\mathbf{y}}$$

sum of squares criterion

$$V = \sum_{i=1}^n e_i^2 = \mathbf{e}^T \cdot \mathbf{e}$$

# Curve-Fitting

## Multilinear regression

### Example

Density of NaCl Aqueous Solutions					
		Temperature			
		0 °C	10 °C	25 °C	40 °C
Wt % NaCl	1	1.00747	1.00707	1.00409	0.99908
	2	1.01509	1.01442	1.01112	1.00593
	4	1.03038	1.02920	1.02530	1.01977
	8	1.06121	1.05907	1.05412	1.04798
	12	1.09244	1.08946	1.08365	1.07699
	16	1.12419	1.12056	1.11401	1.10688
	20	1.15663	1.15254	1.14533	1.13774
	24	1.18999	1.18557	1.17776	1.16971
	26	1.20709	1.20254	1.19443	1.18614

from *Perry's Chemical Engineer's Handbook*,  
Green and Southard, Ed., 9th Ed., p. 2-103.

### Model

$$\rho = \beta_0 + \beta_1 w + \beta_2 T + \beta_3 w^2 + \beta_4 T^2 + \beta_5 wT$$

# Curve-Fitting

## Multilinear regression using vector-matrix calculations

Prepare the dataset

```
import numpy as np
import matplotlib.pyplot as plt

wtpct = np.array([1.,2.,4.,8.,12.,16.,20.,24.,26.])
T = np.array([0.,10.,25.,40.,60.,80.,100.])
Den = np.loadtxt('SaltDensity.csv',delimiter=',',unpack=True)

n1 = len(wtpct)
n2 = len(T)

wn = np.zeros(n1*n2)
Tn = np.zeros(n1*n2)
y = np.reshape(Den,(n1*n2,1),order='F')

for i in range(n1):
    for j in range(n2):
        wn[i*n2+j] = wtpct[i]
        Tn[i*n2+j] = T[j]
```

**MultilinearRegression.py**

# Curve-Fitting

Multilinear regression using vector-matrix calculations

Form the X matrix and carry out the regression calculations

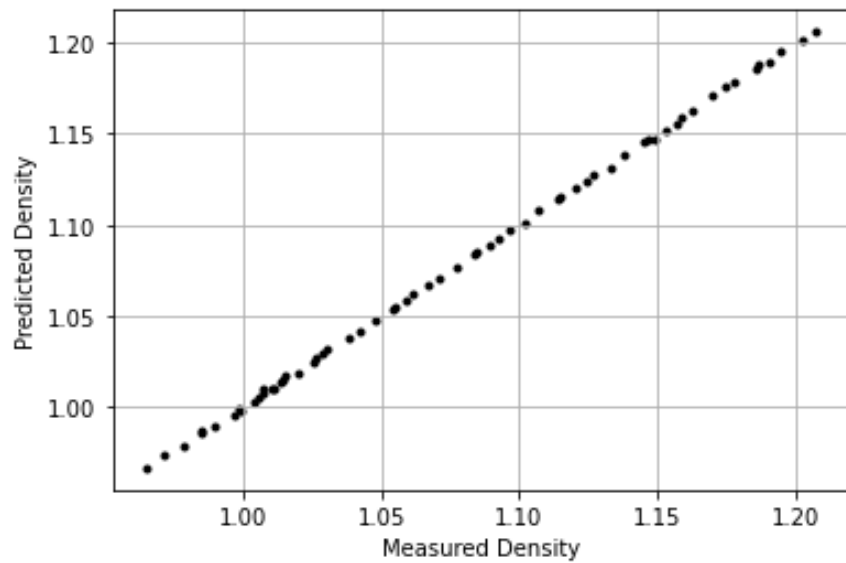
```
X = np.column_stack((np.ones(n), wn, Tn, wn**2, Tn**2, wn*Tn))
Xt = np.transpose(X)
XtX = Xt.dot(X)
Xty = Xt.dot(y)
b = np.linalg.solve(XtX,Xty)
print(b)
```

Model coefficients from left to right

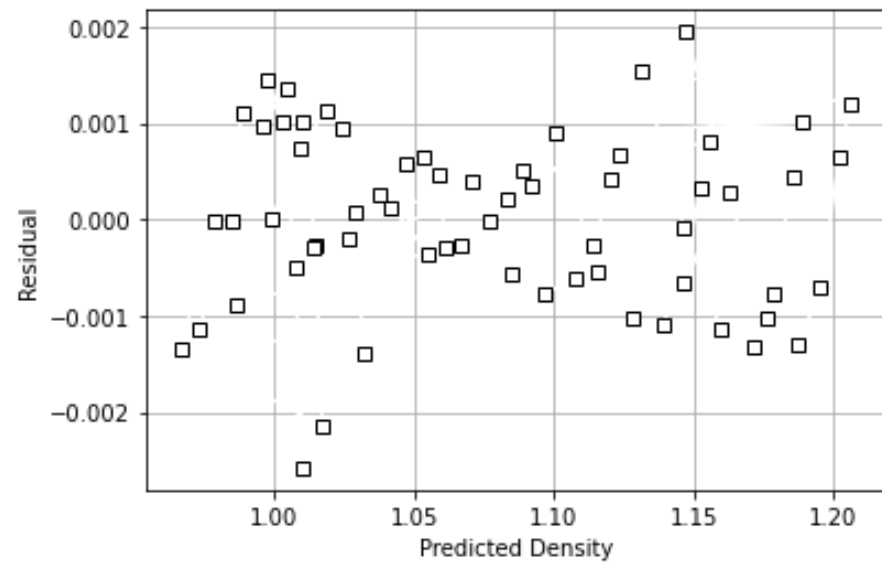
```
[[ 1.00291293e+00]
 [ 7.10909423e-03]
 [-2.21137527e-04]
 [ 2.68240071e-05]
 [-2.08788409e-06]
 [-6.01447754e-06]]
```

# Curve-Fitting

Multilinear regression using vector-matrix calculations



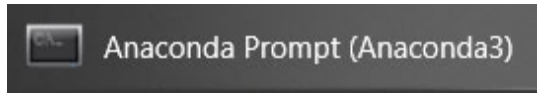
Very close to perfect agreement



Model appears to be adequate

# Curve-Fitting

## Multilinear regression using the **statsmodels** module



```
conda install -c conda-forge statsmodels
```

```
import numpy as np
import matplotlib.pyplot as plt
import statsmodels.api as ssm

wtpct = np.array([1.,2.,4.,8.,12.,16.,20.,24.,26.])
T = np.array([0.,10.,25.,40.,60.,80.,100.])
Den = np.loadtxt('SaltDensity.csv',delimiter=',',unpack=True)

n1 = len(wtpct)
n2 = len(T)
n = n1*n2

wn = np.zeros(n)
Tn = np.zeros(n)
y = np.reshape(Den, (n,1),order='F')
y = y.flatten()

for i in range(n1):
    for j in range(n2):
        wn[i*n2+j] = wtpct[i]
        Tn[i*n2+j] = T[j]
```

### MultiLinearRegressionStatsModels.py



# Curve-Fitting

## Multilinear regression using the **statsmodels** module

```
X = np.column_stack((np.ones(n), wn, Tn, wn**2, Tn**2, wn*Tn))

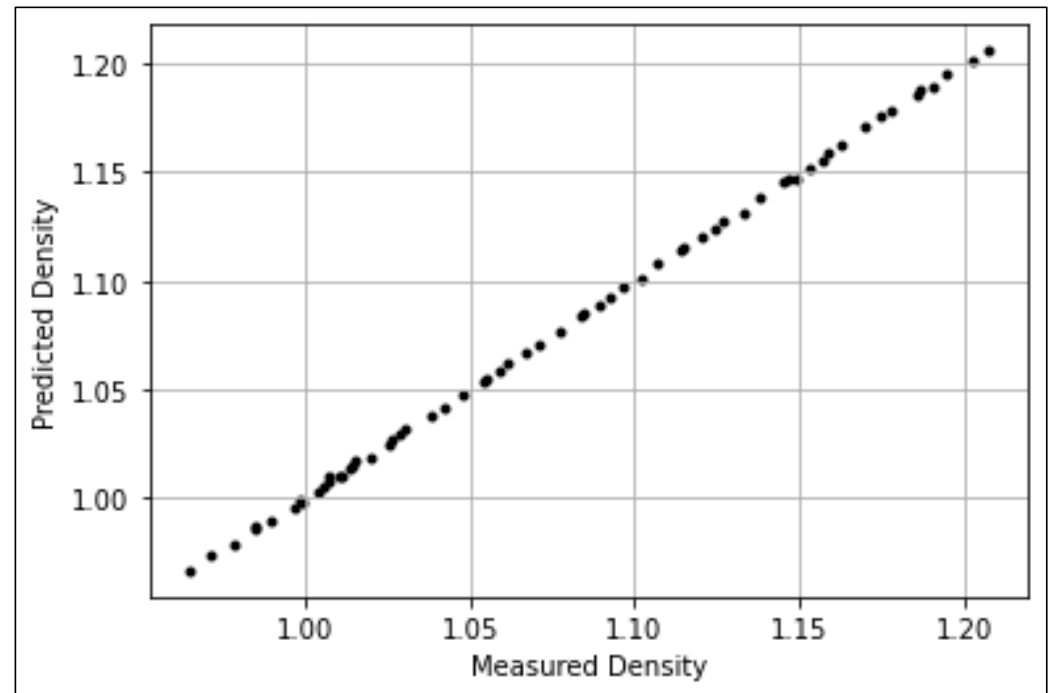
mod = ssm.OLS(y,X)
result = mod.fit()

print(result.params)
print('\n',result.summary())

b = result.params
yp = X.dot(b)
resid = y - yp

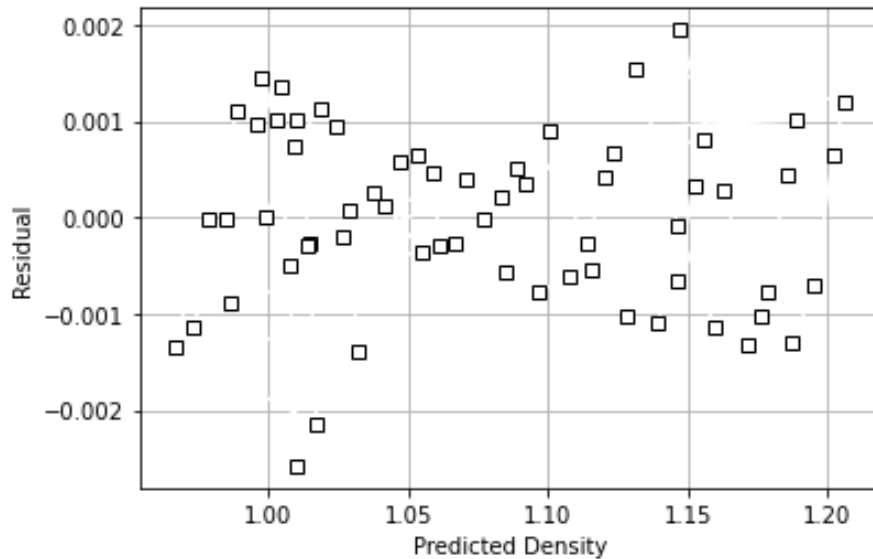
plt.scatter(y,yp,marker='.',c='k')
plt.grid()
plt.xlabel('Measured Density')
plt.ylabel('Predicted Density')

plt.figure()
plt.plot(yp,resid,c='w',marker='s',mec='k',mfc='w')
plt.grid()
plt.xlabel('Predicted Density')
plt.ylabel('Residual')
```



# Curve-Fitting

## Multilinear regression using the **statsmodels** module



```
[ 1.00291293e+00  7.10909423e-03 -2.21137527e-04  2.68240071e-05
 -2.08788409e-06 -6.01447754e-06]
```

### OLS Regression Results

```
=====
Dep. Variable:          y  R-squared:          1.000
Model:                 OLS  Adj. R-squared:       1.000
Method:                Least Squares  F-statistic:      6.543e+04
Date:                  Thu, 01 Jun 2023  Prob (F-statistic):  9.06e-106
Time:                  16:22:12  Log-Likelihood:    351.10
No. Observations:      63  AIC:              -690.2
Df Residuals:          57  BIC:              -677.3
Df Model:              5
Covariance Type:       nonrobust
=====
```

```
=====
              coef  std err      t  P>|t|  [0.025  0.975]
-----
const         1.0029   0.000  2330.561  0.000   1.002   1.004
x1             0.0071  5.97e-05  119.130  0.000   0.007   0.007
x2            -0.0002  1.4e-05  -15.774  0.000  -0.000  -0.000
x3             2.682e-05  2.09e-06  12.829  0.000  2.26e-05  3.1e-05
x4            -2.088e-06  1.28e-07  -16.352  0.000  -2.34e-06  -1.83e-06
x5            -6.014e-06  3.99e-07  -15.057  0.000  -6.81e-06  -5.21e-06
=====
```

```
=====
Omnibus:          1.424  Durbin-Watson:      0.825
Prob(Omnibus):    0.491  Jarque-Bera (JB):    1.287
Skew:             -0.342  Prob(JB):            0.525
Kurtosis:         2.850  Cond. No.            1.70e+04
=====
```

#### Notes:

- [1] Standard Errors assume that the covariance matrix of the errors is correctly specified.
- [2] The condition number is large, 1.7e+04. This might indicate that there are strong multicollinearity or other numerical problems.

# Curve-Fitting

Nonlinear regression

Model:  $y = f(\mathbf{x}, \boldsymbol{\beta})$     Dataset:  $\{y_i, x_{1i}, \dots, x_{mi}, i = 1, \dots, n\}$

$$\mathbf{f}(\mathbf{x}, \boldsymbol{\beta}) = \begin{bmatrix} f(\mathbf{x}_1) \\ f(\mathbf{x}_2) \\ \vdots \\ f(\mathbf{x}_n) \end{bmatrix}$$

$\mathbf{e} = \mathbf{y} - \mathbf{f}(\mathbf{x}, \hat{\boldsymbol{\beta}})$      $\min_{\hat{\boldsymbol{\beta}}} \mathbf{e}^T \mathbf{e}$     using an optimization routine

# Curve-Fitting Nonlinear regression

Example: fitting the Antoine equation to vapor pressure data

$$\log_{10} P_V = A - \frac{B}{C + T}$$

**Vapor Pressure of  
95%(wt) Sulfuric Acid  
Aqueous Solution**

Temperature (degC)	Vapor Pressure (torr)
35	0.0015
40	0.00235
45	0.0037
50	0.0058
55	0.00877
60	0.0133
65	0.0196
70	0.0288
75	0.0415
80	0.0606
85	0.0879
90	0.123
95	0.172
100	0.237
105	0.321
110	0.437
115	0.59
120	0.788
125	1.07
130	1.42
135	1.87
140	2.4
145	3.11
150	4.02
155	5.13
160	6.47
165	8.39
170	10.3
175	12.9
180	15.9
185	20.2
190	24.8
195	30.7
200	36.7
205	45.3
210	55
215	66.9
220	79.8
225	95.5
230	115
235	137
240	164
245	193
250	229
255	268
260	314
265	363
270	430
275	500
280	580
285	682
290	790

# Curve-Fitting Nonlinear regression

Example: fitting the Antoine equation to vapor pressure data

```
import numpy as np
import matplotlib.pyplot as plt
from scipy.optimize import minimize

def SSE(x,T,LVP):
    A = x[0]
    B = x[1]
    C = x[2]
    LVPM = A - B / (C + T)
    VPerr = LVP - LVPM
    return VPerr.dot(VPerr)

T,VP = np.loadtxt('H2SO4VaporPressure.txt',unpack=True)
LVP = np.log10(VP)

A = 10 ; B = 2000 ; C = 250
x0 = [ A, B, C ]

result = minimize(SSE,x0,args=(T,LVP))
print(result)

A = result.x[0]
B = result.x[1]
C = result.x[2]

Tplot = np.linspace(np.min(T),np.max(T),100)
LVplot = A - B / (C + Tplot)

plt.plot(T,LVP,c='w',marker='s',mec='k',mfc='w',label='data')
plt.plot(Tplot,LVplot,c='k',label='Antoine Eqn')
plt.grid()
plt.xlabel('Temperature - degC')
plt.ylabel('log10(Vapor Pressure)')
plt.legend()
```

**NonlinearRegression.py**

# Curve-Fitting Nonlinear regression

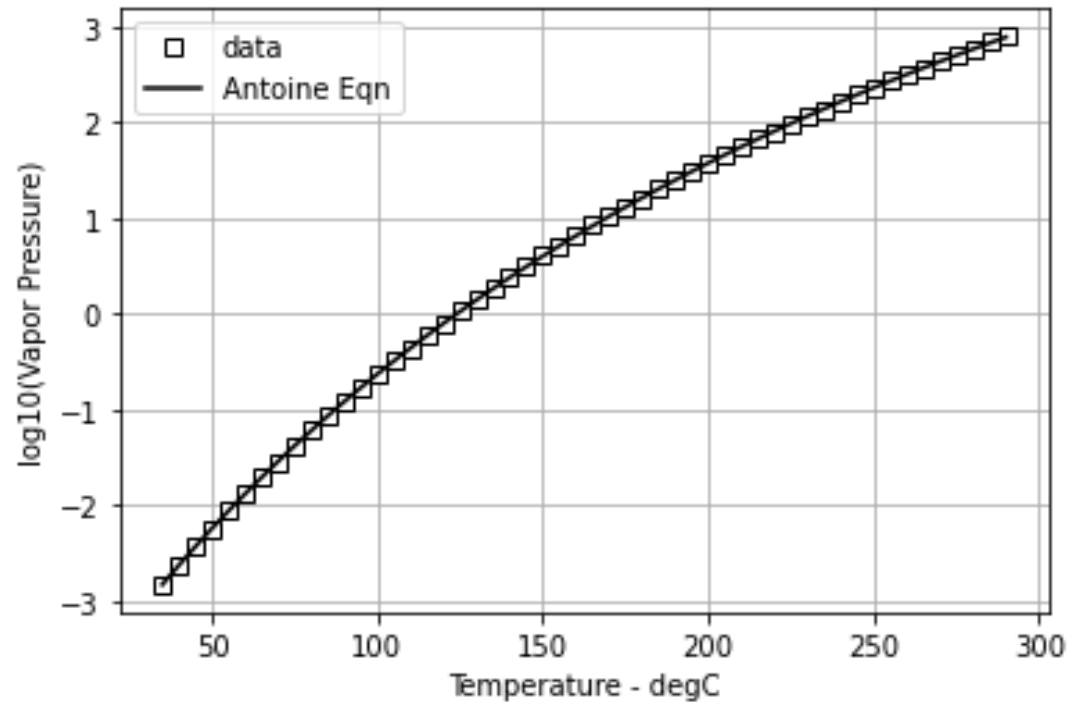
Example: fitting the Antoine equation to vapor pressure data

```
message: Optimization terminated successfully
success: True
status: 0
fun: 0.0009705402527890349
x: [ 9.806e+00  3.902e+03  2.739e+02]
nit: 51
jac: [-4.518e-06  5.178e-08 -9.486e-07]
hess_inv: [[ 1.222e+01  1.022e+04  5.318e+02]
           [ 1.022e+04  8.589e+06  4.488e+05]
           [ 5.318e+02  4.488e+05  2.354e+04]]
nfev: 232
njev: 58
```

A = 9.806

B = 3902

C = 273.9



Reference:

## Applied Numerical Methods with Python

Steven C. Chapra

David E. Clough

McGraw-Hill, 2022

What's Next?

### Python Bootcamps 1, 2 and 3

- ✓ 1: Getting up to speed with Python
- ✓ 2: Learning to use Python to solve typical problem scenarios
- 3: Detailed modeling of packed-bed and plug-flow reactors



"Prof. Clough, may I be excused? My brain is full."